

Module 9: Packed beds

Lecture 31: Examples on pressure drop calculations

 examples on packed bed calculation

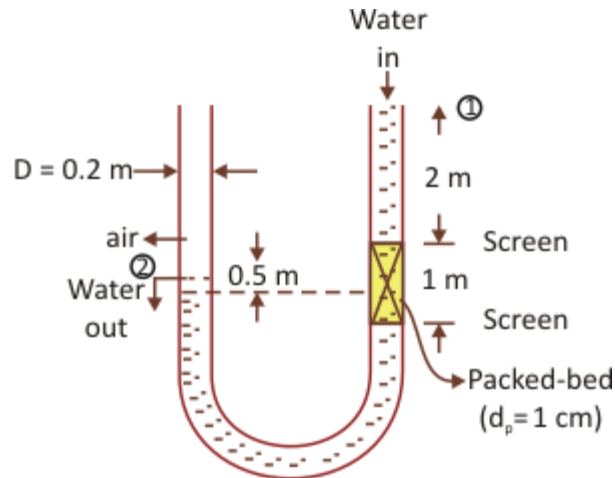
 **Previous** **Next** 

Module 9: Packed beds

Lecture 31: Examples on pressure drop calculations

Examples on packed bed calculation

Shown below is the schematic of a 0.2 m ID U-glass tube to a depth of 1 m with 10,000 spherical quartz $\rho = 7800 \text{ kg/m}^3$ particles ($d_p = 1 \text{ cm}$). What will be water ($\rho = 1000 \text{ kg/m}^3, \mu = 0.001 \text{ kg/cm-s}$) flow rate required through the bed if the water level is kept 2 m above the top of the packed bed? The particles are held between the two sieves.



(Fig. 31a)

Solution: Apply the ME balance between (1) and (2) incorporating frictional loss in the packed-bed:

$$J/kg : g\Delta z + \Delta \left(\frac{u^2}{2} \right) + \left(\frac{\Delta p}{\rho} \right) + w_p + w_s = 0$$

Here, $u_1 = u_2 = \text{constant}$; $P_1 = P_2 = 1 \text{ atm}$, $w_p = 0$

Module 9: Packed beds

Lecture 31: Examples on pressure drop calculations

Therefore, $g(Z_1 - Z_2) = w_s$

Or,

$$w_s = 9.8 \times (2 + 0.5) = 24.5 \text{ J/kg}$$

$$w_s \equiv \frac{150 (1 - \epsilon)^2 \mu_f v_o L}{\epsilon^3 d_p^2 \phi_s^2 \rho_f} + \frac{1.75 (1 - \epsilon) v_o^2 L}{\epsilon^3 d_p \phi_s}$$

$$\epsilon = \text{bed - porosity} = \frac{V_{\text{Packed-bed}} - V_{\text{particles}}}{V_{\text{Packed-bed}}}$$

$$= \frac{\pi/4 (0.2)^2 \times 1 - 1000 \frac{4}{3} \pi \left(\frac{0.01}{2}\right)^3}{\pi/4 (0.2)^2 \times 1}$$

$$= 0.833$$

$\phi_s = 1$: Spherical particle

$$d_p = 0.01 \text{ m}$$

$$\mu_f = 0.001 \text{ kg/m-s}$$

$$\rho_f = 1000 \text{ kg/m}^3$$

Substitute,

$$24.5 = \frac{150 \times (1 - 0.833)^2 \times 0.001 \times V_o \times 1}{0.833^2 \times (0.01)^2 \times 1^2 \times 1000} + \frac{1.75(1 - 0.833)V_o^2 \times 1}{0.833^2 \times 0.01}$$

Or

$$24.5 = 0.0723 v_o + 50.56 V_o^2$$

$$V_o = 0.69 \text{ m/s}$$

$$Q = \pi/4 D^2 \times V_o = \frac{\pi/4 (0.2)^2 \times 0.69}{1000} \times 10^6 \times 60$$

$$= 1300 \text{ liters per min.}$$

◀ Previous Next ▶

Module 9: Packed beds

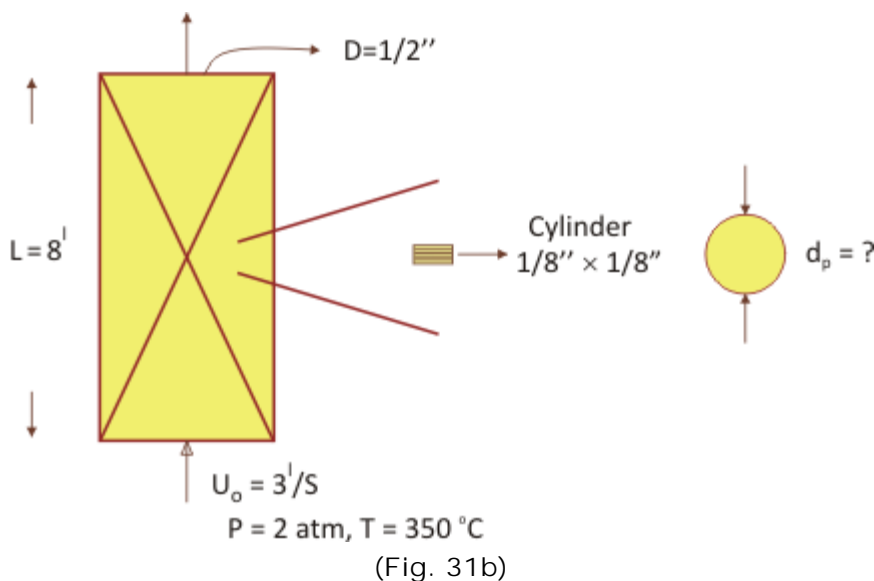
Lecture 31: Examples on pressure drop calculations

Example 2:

Partial oxidation is carried out by passing air with 1.2 mol percent hydrocarbons through $\frac{1}{2}$ ID tube packed with 8' of $1/8'' \times 1/8''$ cylindrical catalyst pellets. The air enters at 350°C and 2.0 atm with a superficial velocity of 3 ft/s. Calculate the pressure-drop because of friction.

Assume, $\epsilon = 0.40, \mu(\text{air}) = 16 \times 10^{-5} \frac{\text{gm} \cdot \text{cm}}{\text{s}}$.

Answer:



First calculate ϕ_s of the packed material (pellet).

$$\phi_s \equiv \frac{4\pi \left(\frac{d_p}{2}\right)^2}{\left(\pi d^2 + 2 \frac{\pi d^2}{4}\right)}; d_p \equiv \text{diameter of a sphere having the same volume as that of}$$

cylindrical pellet.

$$= \frac{2}{3} \left(\frac{d_p}{d}\right)^2, \text{ surface area of cylindrical pellet, } d = \text{length or diameter of the pellet. And,}$$

$$\frac{4}{3} \pi \left(\frac{d_p}{d}\right)^3 = \frac{\pi d^2 \cdot d}{4} \quad (= \text{volume of the pellet})$$

$$d^3 = \frac{2}{3} d_p^3 \Rightarrow \left(\frac{d_p}{d}\right)^3 = \frac{3}{2} \Rightarrow d_p = 0.363 \text{ cm}$$

◀ Previous Next ▶

Module 9: Packed beds

Lecture 31: Examples on pressure drop calculations

Therefore,

$$\phi_s = \frac{2}{3} \left(\frac{3}{2} \right)^{2/3} = 0.667 (1.5)^{2/3}$$

$$\text{Calculate } Re_p = \frac{v_o d_p \rho_f}{\mu_f} = \frac{(3 \times 12 \times 2.54) \times 0.363 \times 0.0011}{16 \times 10^{-5}}$$

$$\left(\rho_f = \frac{PM}{RT} = \frac{2 \times 28 \times 10^{-3}}{623 \times .082 \times 10^{-6}} \right)$$

$$= 228$$

Therefore, both viscous and inertial terms are important and Ergun's equation should be written as:

$$\frac{\Delta p}{\rho_f} = \frac{150 (1 - \epsilon)^2 \mu_f v_o L}{\epsilon^3 d_p^2 \phi_s^2 \rho_f} + \frac{1.75 (1 - \epsilon) v_o^2 L}{\epsilon^3 d_p \phi_s}$$

$$\text{Substitute, } \Delta p = \frac{150 \times (1 - 0.4)^2 \times 16 \times 10^{-5} \times (3 \times 12 \times 2.54) \times (8 \times 12 \times 2.54)}{0.4^3 \times .363^2 \times 0.667 (1.5)^{2/3}}$$

$$+ \frac{1.75 \times (1 - 0.4) \times (3 \times 12 \times 2.54)^2 \times (8 \times 12 \times 2.54)}{0.4^3 \times 0.363 \times 0.667 (1.5)^{2/3}}$$

$$= 124225 \text{ Dyne/cm}^2$$

(Note: It is assumed that the variation in gas-velocity because of variation in pressure is negligible.

v_o and ρ_f are calculated at the inlet conditions).

◀ Previous Next ▶