

Module 1: Introduction

Lecture 2: Rate of strain, Non-Newtonian fluid

☰ Rate of strain or strain rate or deformation rate

☰ Revert to Newton's' 1st Law of viscosity

☰ Non–Newtonian fluid (Introduction)

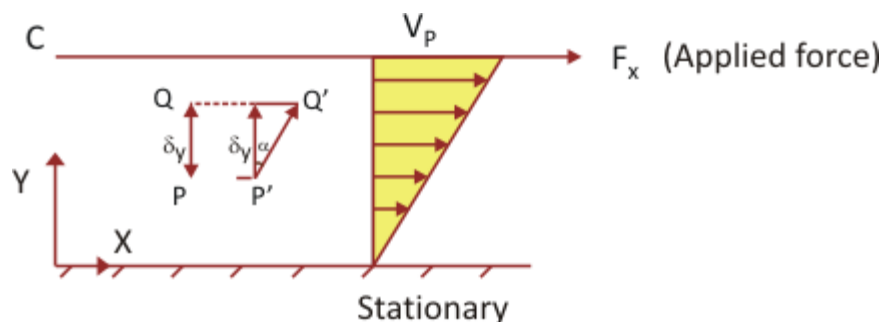
◀ Previous Next ▶

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Rate of strain or strain rate or deformation rate

- Consider the flow of a fluid between two solid plates under the influence/application of a force in X-direction on the top plate.
- The bottom plate is stationary. The top surface of the fluid has a constant velocity, V_p
- It has been experimentally demonstrated that in such case 'no – slip' condition exists. The fluid layer in contact with the bottom solid surface will be stationary, whereas the layer in contact with the top solid surface will move at the velocity, V_p . The velocity profiles between the plates will be linear.



(Fig. 2a)

As shown in figure 2a, the fluid element QP of length δy at time $t = 0$ becomes $Q'P'$ at time, $t = \delta t$. Difference in the displacements of Q and P to Q' and P' , respectively, is because of different velocity at Q and P .

◀ Previous Next ▶

Module 1: Introduction

Lecture 2: Rate of strain, Non-Newtonian fluid

Therefore,

$$PP' = V_x \delta t$$

$$QQ' = \left(V_x + \frac{dV_x}{dy} \cdot \delta y \right) \cdot \delta t$$

Difference in the length of the segments QP and $Q'P'$

$$= (Q'P' - QP)$$

$$\simeq \left(V_x + \frac{dV_x}{dy} \cdot \delta y \right) \delta t - V_x \delta t \text{ elongation}$$

$$= \frac{dV_x}{dy} \cdot \delta y \cdot \delta t$$

$$\left. \begin{array}{l} \text{Rate of deformation} \\ \text{or strain rate} \\ \text{or deformation rate} \end{array} \right\} \equiv \left(\frac{\text{Elongation of the Fluid Element}}{\text{Original length}} \right) \times \frac{1}{\delta t}$$

$$= \frac{\frac{dV_x}{dy} \cdot \delta y \cdot \delta t}{\delta y \cdot \delta t}$$

$$= \left(\frac{dV_x}{dy} \right)$$

◀ Previous Next ▶

Module 1: Introduction

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Revert to Newton's' 1st Law of viscosity

τ	$= \mu$	$\frac{dv_x}{dy}$
↓	↓	↓
shear stress	viscosity	strain rate

It has been shown that

$$\mu \equiv \mu \left(T, \frac{dv_x}{dy} \right) \text{ in general.}$$

For Newtonian fluid (water, air, glycerin, etc): $\mu \equiv \mu(T)$. For liquids, viscosity increases with temperature. For gases, viscosity decreases with temperature.

For Non-Newtonian fluid: $\mu \equiv \mu \left(\frac{dv_x}{dy} \right)^n$. Examples are sugar solution and polymers.

Therefore, General expression for shear stress

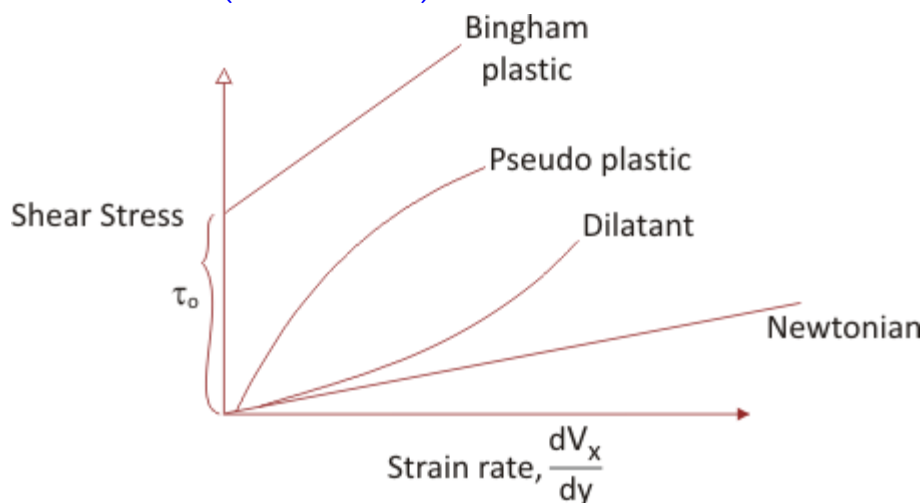
$$\tau = m \left| \frac{dv_x}{dy} \right|^{n-1} \left(\frac{dv_x}{dy} \right)$$

where, m is the flow consistency and n is the flow behaviour index.

- Also known as power-Law model
- $m \left| \frac{dv_x}{dy} \right|^{n-1}$ also known as apparent viscosity

◀ Previous Next ▶

Non – Newtonian fluid (Introduction)



(Fig 2b)

- Newtonian: $\tau = \mu \frac{dV_x}{dy}$: air, water, glycerin
- Bingham Plastic: $\tau = \tau_0 + \mu \frac{dV_x}{dy}$: toothpaste
 (Fluid does not move or deform till there is a critical stress)
- Dilatant: $\tau = K \left(\frac{dV_x}{dy} \right)^n, n > 1$: starch or sand suspension
 or shear thickening fluid
 (Fluid starts 'thickening' with increase in its apparent viscosity)
- Pseudo plastic: $\tau = K \left(\frac{dV_x}{dy} \right)^n, n < 1$: paint or shear thinning fluid
 (Fluid starts 'thinning' with decrease in its apparent viscosity)

Note:

1. For some non-Newtonians fluids, viscosity or apparent viscosity may be time- dependent. Such fluids are also called 'Memory' fluids.
2. Rheology is a science of studying flow and behavior of polymeric fluids.