

Unit 7 - Week 5

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Assignment 5

The due date for submitting this assignment has passed.
As per our records you have not submitted this assignment.

Due on 2020-10-21, 23:59 IST.

- 1) Ion with positive charge is known as? 1 point
- a) Cation
b) Anion
c) Positron
d) Negatron
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
a)
- 2) Cation exchange capacity (CEC) of a soil is the measure of quantity of _____ charged sites on soil surfaces that can retain _____ charged ions by electrostatic forces? 1 point
- a) Positive, negative
b) Negative, positive
c) Positive, positive
d) Negative, negative
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
b)
- 3) The buffered methods of measuring CEC will determine _____? 1 point
- a) Effective CEC
b) True CEC
c) Potential or maximum CEC
d) All of the above
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
c)
- 4) The most widely used methods of determining CEC are? 1 point
- a) Distillation method
b) Centrifuge method
c) Both a and b
d) None of the above
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
c)
- 5) Ion Exchange Capacity (IEC) is expressed as _____? 1 point
- (CEC = Cation Exchange Capacity, AEC = Anion Exchange Capacity)
- a) IEC = CEC/AEC
b) IEC = CEC+AEC
c) IEC = CEC*AEC
d) IEC = CEC - AEC
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
b)
- 6) Which of the following is True with respect to AEC and pH? 1 point
- a) AEC is unaffected by pH
b) AEC increases with increase in pH
c) AEC decreases with increase in pH
d) AEC first increases then decreases with an increase in pH
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
c)
- 7) Hydrogen bond is found in which of the following clay mineral? 1 point
- a) Kaolinite
b) Montmorillonite
c) Vermiculite
d) Beidelite
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
a)
- 8) Which of the following has the highest CEC? 1 point
- a) Kaolinite
b) Smectite
c) Vermiculite
d) Humus
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
d)
- 9) Percent base saturation is the % of CEC occupied by _____ ion/ions. 1 point
- a) Al^{3+} and H^+
b) Ca^{2+} , Mg^{2+} , Na^+ , K^+
c) Cr^{3+}
d) All of the above
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
b)
- 10) Low CEC soils are _____. 1 point
- a) Deficient in potassium and magnesium
b) Have low water holding capacity
c) Both a and b
d) None of the above
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
c)
- 11) How many molc of K^+ does it take to replace 12 molc of Ca^{2+} ? 1 point
- a) 0.5
b) 1
c) 2
d) 12
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
d)
- 12) What will be the CEC of a soil which has 5 cmolc/kg of H^+ , 6 cmolc/kg of Ca^{2+} , 3 cmolc/kg of Mg^{2+} , 2 cmolc/kg of Na^+ and 1 cmolc/kg of K^+ ? 1 point
- a) 17 cmolc/kg
b) 12 cmolc/kg
c) 6 cmolc/kg
d) 3 cmolc/kg
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
a)
- 13) As per Kerr, the selectivity coefficient for Ca-Mg exchange complex _____? 1 point
- a) Increased with change in exchanger composition
b) Relatively constant as exchanger composition changed
c) Decreased with change in exchanger composition
d) None of the above
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
b)
- 14) When a mixture of cations A and B is ideal, the exchanger phase activity coefficients for them will be given by _____. 1 point
- a) $f_A = f_B = 0.1$
b) $f_A = f_B = 1$
c) $f_A = f_B = 10$
d) $f_A = f_B = 100$
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
b)
- 15) The strength of adsorption of common cations on most colloids is in the order of? 1 point
- a) $Al^{3+} > Ca^{2+} > Mg^{2+} > K^+ = NH_4^+ > Na^+$
b) $Al^{3+} < Ca^{2+} > Mg^{2+} > K^+ = NH_4^+ > Na^+$
c) $Al^{3+} < Ca^{2+} < Mg^{2+} < K^+ = NH_4^+ < Na^+$
d) $Al^{3+} > Ca^{2+} < Mg^{2+} > K^+ = NH_4^+ < Na^+$
- a)
 b)
 c)
 d)
- No, the answer is incorrect.
Score: 0
Accepted Answers:
a)