

Unit 9 - Week 8 : Third Generation Solar Cells

Course outline

How does an NPTEL online course work?

Week 1 : Introduction and Solar radiation fundamentals

Week 2 : Basic physics of semiconductors

Week 3 : Carrier transport, generation and recombination in semiconductors

Week 4 : Semiconductor junctions

Week 5 : Essential characteristics of solar photovoltaic devices

Week 6 : First Generation Solar Cells

Week 7 : Second Generation Solar Cells

Week 8 : Third Generation Solar Cells

Lecture 36 : Generation II Technologies: CIGS Solar Cells

Lecture 37 : Generation II Technologies: CIGS and Multijunction Solar Cells

Lecture 38 : Generation III Technologies : Organic Solar Cells

Lecture 39 : Generation III Technologies: Organic Solar Cells

Lecture 40 : Generation III Technologies: Organic and Dye Sensitized Solar Cells

Lecture 41 : Generation III Technologies: Perovskite and CZTS Solar Cells

Quiz : Assignment 8

Solar Photovoltaics: Principles, Technologies and Materials: Week 8 Feedback

Assignment 8 Solution

Text Transcripts

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Assignment 8

The due date for submitting this assignment has passed.
As per our records you have not submitted this assignment.

Due on 2020-03-25, 23:59 IST.

1) In CIGS solar cells, what is the starting source of sulphur for window layer deposition? 1 point

- CdSO₄
 H₂S
 SC(NH₂)₂
 S

No, the answer is incorrect.
Score: 0

Accepted Answers:

SC(NH₂)₂

2) CuInSe₂ (E_g=1.01 eV) is used as an absorber layer in solar cells. It is doped with Ga which occupies In site. For atomic ratio of In:Ga = 3:1, what is likely to happen? 1 point

- Band gap will increase to 1.04eV.
 Band gap will decrease to 0.95 eV.
 Crystal structure changes.
 Short circuit current will decrease.

No, the answer is incorrect.
Score: 0

Accepted Answers:

Band gap will increase to 1.04eV.

Short circuit current will decrease.

3) In CIGS solar cells, which of the following defects are responsible for increased recombination: 1 point

- Copper vacancies
 Indium atoms substituting copper sites
 Selenium vacancies
 Copper atom substituting indium sites

No, the answer is incorrect.
Score: 0

Accepted Answers:

Copper atom substituting indium sites

4) CIGS solar cells use soda lime glass as substrates. What could the possible reasons? 1 point

- Soda lime glass suppresses the impurity formation in CIGS.
 Na in Soda lime glass substrates improves surface passivation of the films.
 Soda lime glass substrates improve crystallographic texture of CIGS
 Soda lime glass substrates reduce thermal stresses in CIGS films.

No, the answer is incorrect.
Score: 0

Accepted Answers:

Na in Soda lime glass substrates improves surface passivation of the films.

Soda lime glass substrates reduce thermal stresses in CIGS films.

5) What are the oxides that can be used as transparent electrodes in solar cells? 1 point

- Pure zinc oxide
 Indium doped Tin Oxide
 Aluminum doped Zinc Oxide
 Pure Tin Oxide

No, the answer is incorrect.
Score: 0

Accepted Answers:

Indium doped Tin Oxide

Aluminum doped Zinc Oxide

6) Which of the following factors play an important role in improving efficiency of the Organic solar cells? 1 point

- Using bulk heterojunction configuration
 High exciton binding energy
 Absorber thickness of the order of 1 micron
 Use of low bandgap polymers

No, the answer is incorrect.
Score: 0

Accepted Answers:

Using bulk heterojunction configuration

7) Which of the following is correct for organic solar cells? 1 point

- Donor is p-type
 Acceptor is p-type
 Space charge region on donor side is smaller than on the acceptor side
 Acceptor concentration is very large in comparison to donor

No, the answer is incorrect.
Score: 0

Accepted Answers:

Donor is p-type

Space charge region on donor side is smaller than on the acceptor side

8) In comparison to inorganic semiconductor such as Si, Organic semiconductors possess which of the following attributes? 1 point

- Lower light absorption
 Lower exciton diffusion length
 Lower bandgap
 Lower carrier mobilities

No, the answer is incorrect.
Score: 0

Accepted Answers:

Lower exciton diffusion length

Lower carrier mobilities

9) Performance of DSSC solar cells can be improved by: 1 point

- Using dyes having broad absorption
 Using a metallic photosensitizer
 Using wide bandgap n-type photosensitizer
 Using counter electrode having lower catalytic activity

No, the answer is incorrect.
Score: 0

Accepted Answers:

Using dyes having broad absorption

Using wide bandgap n-type photosensitizer

10) In a perovskite solar cells, which of the following is correct: 1 point

- Absorber layer is direct band gap material.
 Exciton diffusion lengths are below 100 nm.
 Absorber layer is 100 nm thick
 Absorber layer is a perovskite-structured material.

No, the answer is incorrect.
Score: 0

Accepted Answers:

Absorber layer is direct band gap material.

Absorber layer is a perovskite-structured material.

11) Which of the following is not strictly true for Dye sensitized solar cells (DSSC): 1 point

- Highly ionic conductive electrolytes such as NaI, LiI are dissolved in protic solvents like acetonitrile, propylene carbonate.
 Wide band gap materials with high surface area are used as photoelectrodes.
 Polypyridyl compounds of Ruthenium and osmium are used as photosensitizers.
 Photosensitizers used should have wide band gap and high absorption coefficient.

No, the answer is incorrect.
Score: 0

Accepted Answers:

Highly ionic conductive electrolytes such as NaI, LiI are dissolved in protic solvents like acetonitrile, propylene carbonate.

12) In n-i-p structured Perovskite solar cells: 1 point

- Spiro-oMeTAD acts as an electron transport layer
 Mesoscopic configuration results in better PCE over planar configuration
 TiO₂ is used as electron selective layer
 Al is used as a top contact.

No, the answer is incorrect.
Score: 0

Accepted Answers:

Mesoscopic configuration results in better PCE over planar configuration

TiO₂ is used as electron selective layer

Al is used as a top contact.

13) What is the correct sequence of processes taking place in Dye sensitized solar cells (DSSC)? 1 point

- Photon absorption → excitation of dye molecules → oxidation of photo sensitizer → electrical energy generation → regeneration of dye molecules → recapture of electrons by electrolyte
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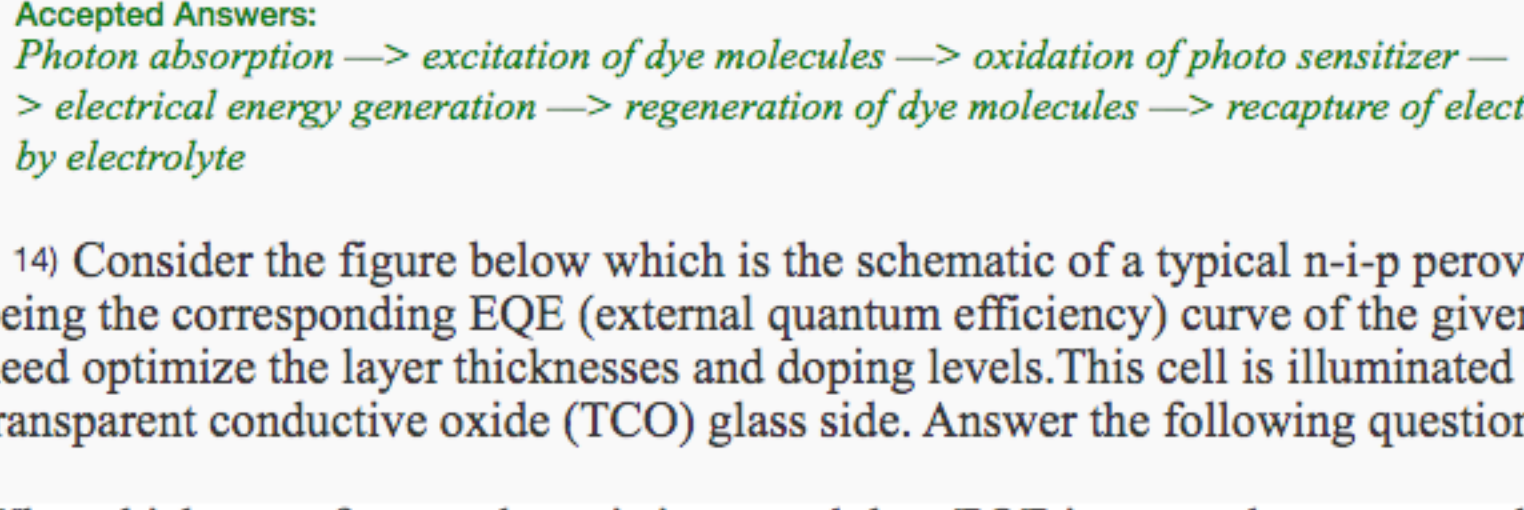
No, the answer is incorrect.
Score: 0

Accepted Answers:

Photon absorption → excitation of dye molecules → oxidation of photo sensitizer → electrical energy generation → regeneration of dye molecules → recapture of electrons by electrolyte

14) Consider the figure below which is the schematic of a typical n-i-p perovskite solar cell with curve labelled (c) being the corresponding EQE (external quantum efficiency) curve of the given cell. For optimization of the device, you need optimize the layer thicknesses and doping levels. This cell is illuminated with AM1.5G irradiation from the transparent conductive oxide (TCO) glass side. Answer the following questions: 1 point

When thickness of p-type layer is increased then EQE is correctly represented by:



- Curve a
 Curve d
 Curve b
 Curve c

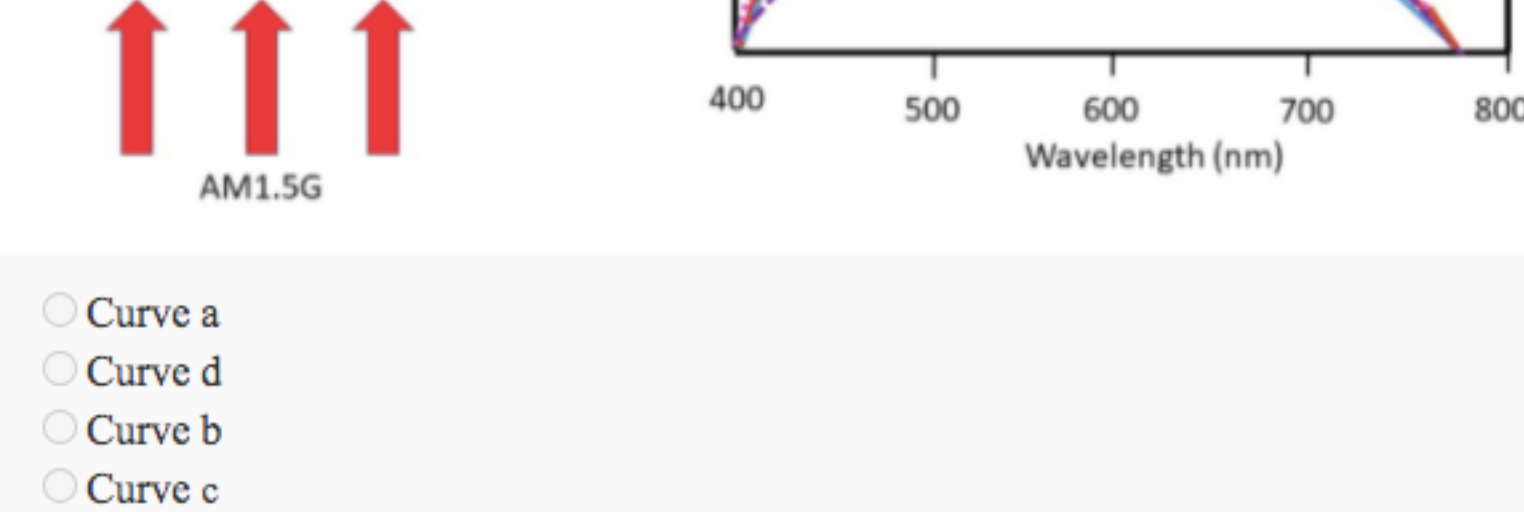
No, the answer is incorrect.
Score: 0

Accepted Answers:

Curve b

15) Consider the figure below which is the schematic of a typical n-i-p perovskite solar cell with curve labelled (c) being the corresponding EQE (external quantum efficiency) curve of the given cell. For optimization of the device, you need optimize the layer thicknesses and doping levels. This cell is illuminated with AM1.5G irradiation from the transparent conductive oxide (TCO) glass side. Answer the following questions: 1 point

If thickness of n-type layer is reduced, then which curve correctly represents the EQE:



- Curve a
 Curve c
 Curve b
 Curve d

No, the answer is incorrect.
Score: 0

Accepted Answers:

Curve d