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reviewer4@nptel.iitm.ac.in ▼

Courses » Introduction to Materials Science and Engineering

Announcements

Course

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Unit 6 - Week 3 - Structure of Solids II

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Course outline

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Assignment 3

The due date for submitting this assignment has passed.

As per our records you have not submitted this
assignment.

Due on 2019-02-20, 23:59 IST.

1) Which of the following stacking sequences of close-packed layers A, B and C will give rise to close-packed structures? **1 point**

P: ABABAB...

Q: ABCABCABC...

R: AABBAABB....

S: ACBACB...

T: ABACABAC...

- ☐ P and Q
- ☐ P, Q and S
- ☐ P, Q, S and T
- ☐ P, Q, R and S

No, the answer is incorrect.

Score: 0

Accepted Answers:

P, Q, S and T

2) Which of the following will always exhibit complete solubility? **1 point**

- ☐ interstitial solid solutions
- ☐ substitutional solid solutions
- ☐ both interstitial and substitutional solid solutions
- ☐ neither substitutional nor interstitial solid solution

No, the answer is incorrect.

Score: 0

Accepted Answers:

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graphite	<input type="radio"/> S	
<input checked="" type="radio"/> 3.8 Structure of diamond	<input type="radio"/> P, Q, R and S	
<input type="radio"/> Quiz : Assignment 3	<input type="radio"/> P and R	
<input type="radio"/> Assignment 3 Solutions	<input type="radio"/> P and S	
Week 4 - Structure of Solids III	No, the answer is incorrect.	
	Score: 0	
	Accepted Answers:	
	<i>P and S</i>	
Week 5 - Defects in Crystalline Solids I	4) What is the effective number of octahedral voids in one unit cell of a CCP crystal?	1 point
	<input type="radio"/> 2	
	<input type="radio"/> 4	
	<input type="radio"/> 6	
	<input type="radio"/> 8	
Week 6 - Defects in Crystalline Solids II	No, the answer is incorrect.	
	Score: 0	
	Accepted Answers:	
	<i>4</i>	
Week 7 - Phase Diagrams I	5) The combinations of which of the following elements are likely to have complete solid solubility based on the Hume-Rothery rules?	1 point
	A (ccp, 1.3 Å radius); B (hcp, 1.3 Å radius); C (hcp, 1.4 Å radius)	
	Assume that the valencies and electronegativities are the same.	
Week 8 - Phase Diagrams II + Diffusion	<input type="radio"/> A and B	
	<input type="radio"/> B and C	
Week 9 - Phase Transformations I	<input type="radio"/> A and C	
	<input type="radio"/> None of the above	
Week 10 - Phase Transformations II + Mechanical Behaviour of Materials I	No, the answer is incorrect.	
	Score: 0	
	Accepted Answers:	
	<i>B and C</i>	
Week 11 - Mechanical Behaviour of Materials II	6) Identify the wrong statement.	1 point
	<input type="radio"/> β-brass is an intermediate solid solution	
	<input type="radio"/> Ordered β-brass is a substitutional solid solution	
	<input type="radio"/> Disordered β-brass is an interstitial solid solution	
	<input type="radio"/> The lattice of ordered β-brass is cubic-P	
Week 12 - Mechanical Behaviour of Materials III + Fracture	No, the answer is incorrect.	
	Score: 0	
	Accepted Answers:	
	<i>Disordered β-brass is an interstitial solid solution</i>	
Interactive Session	7) The crystal structure of diamond is _____ and the lattice is _____.	1 point
	<input type="radio"/> cubic, cubic-P	
	<input type="radio"/> cubic close packed, cubic-F	
	<input type="radio"/> diamond cubic, diamond cubic	
	<input type="radio"/> diamond cubic, cubic-F	

No, the answer is incorrect.

Score: 0

Accepted Answers:

diamond cubic, cubic-F

8) Out of elements A, B and C, identify the ones that can fit inside the voids of a hexagonal close packed structure of a defect-free metal X? ($r_X=1.28 \text{ \AA}$, $r_A=0.25 \text{ \AA}$, $r_B=0.66 \text{ \AA}$, $r_C=0.50 \text{ \AA}$) **1 point**

- ☐ A
- ☐ A, B and C
- ☐ A and C
- ☐ B and C



No, the answer is incorrect.

Score: 0

Accepted Answers:

A and C

9) Which of the following are the centres of octahedral voids in a CCP unit cell? **1 point**
P (1,1,1); Q ($\frac{1}{2}, 0, 0$); R ($\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$); S ($\frac{1}{4}, \frac{1}{4}, \frac{1}{4}$); T ($\frac{1}{2}, \frac{1}{2}, 0$).

- ☐ Q, R and S
- ☐ P and T
- ☐ P, Q, R and T
- ☐ Q and R

No, the answer is incorrect.

Score: 0

Accepted Answers:

Q and R

10) A given CCP crystal has the stacking sequence ABCABC... Which of the following are the closest from the centre of a tetrahedral void located between the layers A and B? **1 point**

P: The centre of a neighbouring tetrahedral void
Q: The centre of a neighbouring octahedral void
R: The centre of a neighbouring atom in the A layer
S: The centre of a neighbouring atom in the B layer
T: The centre of a neighbouring atom in the C layer

- ☐ P and Q
- ☐ R and S
- ☐ R, S and T
- ☐ Q, R and S

No, the answer is incorrect.

Score: 0

Accepted Answers:

Q, R and S

11) Let a be the shortest lattice translation and b be the C-C bond length in graphene. The relationship between a and b is given by _____. **1 point**



$$a = b$$



$$a = \sqrt{3}b$$



$$a = \sqrt{2}b$$



$$a = \sqrt{3}b/2$$

No, the answer is incorrect.

Score: 0

Accepted Answers:

$$a = \sqrt{3}b$$



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