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## Courses » Fluid Inclusions in Minerals: Principles, Methodology, Practice and Applications

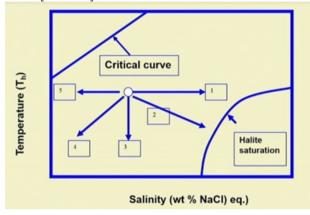
Announcements Course Ask a Question Progress Mentor FAQ

Due on 2018-09-05, 23:59 IST

1. Define salting out effect? [2 marks]

2. What is freezing point depression and how is it different from super cooling? [2 marks]

- 3. Define temperature of homogenization and how is it different from temperature of entrapment? [2 marks] 4. What are the parameters on which density of aqueous fluid inclusion is dependent on? Which will be of higher density; aqueous biphase inclusion at room temperature or at its homogenization temperature condition? Why? [2 marks]
- 5. What are the various possibilities of homogenization in case of an aqueous biphase inclusion? [2 marks]
- 6. What are the various hints which poke towards the metastability of an aqueous biphase inclusions? [2 marks]
- 7. Which field will denote evolution of polyphase inclusion in the attached diagram? What will be the probable explanation for aqueous biphase inclusion without a halite crystal but falling beyond the halite saturation field? [2 marks]



- 8. What are the probable fluid evolution paths (that is 1,2,3,4, and 5 in the above fluid evolution diagram) that a crustal fluid would behave over a time period?[2 marks]
- 9. What happens to a fluid inclusion which is heated beyond the temperature of homogenization? Why one should avoid such fluid inclusions for study?[2 marks]
- 10. What are possible relation that can exist between T<sub>s, NaCl</sub> (temperature of halite dissolution) and T<sub>h</sub> (temperature of homogenization)? Mention the temperature (whether T<sub>s, NaCl</sub> or T<sub>h</sub>) which will be used to deduce salinity and density in each case.
  [2 marks]

## **Your Submission:**

Due Date Exceeded.

As per our records you have not submitted this assignment.



Ministry of Human Resource Development

