

Course outline

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● Week 7 Lecture Material

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Week 7 : Assignment 7

The due date for submitting this assignment has passed.

Due on 2021-09-15, 23:59 IST.

As per our records you have not submitted this assignment.

 1) **An isolated system;** 2 points

- (A) May have insulated boundary.
- (B) Must have both insulated and rigid boundary.
- (C) May have rigid boundary.
- (D) May have either insulated or rigid boundary.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

B)

 2) **A hot water bag is an example of;** 2 points

- (A) Open System
- (B) Isolated System
- (C) Closed System
- (D) Steady State System

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

C)

 3) **The ratio of two extensive properties is;** 2 points

- (A) Extensive properties
- (B) Is always constant
- (C) Is always unit less
- (D) Intensive properties

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

D)

 4) **Pick up the correct statement.** 2 points

- (A) A glass of water is a closed system.
- (B) A closed system is always at steady state.
- (C) An isolated system is always at steady state.
- (D) A pressure cooker is a closed system.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

C)

 5) **Pick up the correct statement.** 2 points

- (A) A closed system can be insulated.
- (B) An isolated system can have pervious or porous boundary.
- (C) An isolated system can have movable or flexible boundary.
- (D) A closed system can never have a rigid boundary.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

A)

 6) **Find out the correct statement.** 2 points

When a glass full of water is heated from bottom;

- (A) The system is initially not at mechanical equilibrium and eventually attains mechanical equilibrium.
- (B) The system is initially not at thermal equilibrium and eventually attains mechanical equilibrium.
- (C) The system is initially not at chemical equilibrium and eventually attains mechanical equilibrium.
- (D) All of the above.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

B)

 7) **The degree of freedom of any point on the phase boundary between liquid and vapour regions of a phase diagram is;** 2 points
The degree of freedom of any point on the phase boundary between liquid and vapour regions of a phase diagram is;

- (A) Always 0.
- (B) Always 2.
- (C) Always 1.
- (D) Always 3.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

C)

 8) **The S (solid) – L (Liquid) phase boundary has a negative slope, is physically substantiated by the fact;** 2 points
The S (solid) – L (Liquid) phase boundary has a negative slope, is physically substantiated by the fact;

- (A) Water has more density at 4 °C.
- (B) Ice is crystalized.
- (C) Melting point of ice reduces with increase in pressure.
- (D) Melting point of ice increase with increase in pressure.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

C)

 9) **Pick up the correct statement.** 2 points

A mixture of two liquids having two distinct boiling point will have;

- (A) A unique boiling point which is lower than the boiling point of the lower boiling liquid.
- (B) A unique boiling point which is higher than the boiling point of the higher boiling liquid.
- (C) Will have a boiling range and not a unique boiling point.
- (D) Will have unique boiling point which will be between the boiling point of the two liquids.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

B)

 10) **For materials the sublines** 2 points
For materials the sublines

- (A) The critical temperature is lower than the room temperature.
- (B) The triple point temperature is lower than the room temperature.
- (C) The triple point temperature is higher than the room temperature.
- (D) The critical temperature is higher than the room temperature.

- A)
- B)
- C)
- D)

No, the answer is incorrect.
Score: 0
Accepted Answers:

C)