#### Nano structured materials-synthesis, properties, self assembly and applications by Prof. A.K. Ganguli, Chemistry Department, IIT Delhi, New Delhi.

### Module 2, Lecture 10: Vapour-Liquid-solid method

#### Problem:

- 1. In VLS technique, the catalyst occurs as \_\_\_\_\_\_ on the substrate. (liquid droplet)
- 2. The growing material precipitates at \_\_\_\_\_\_. (Interface between the substrate and liquid).
- 3. Which type of growth is achieved by VLS technique? (1D growth)
- 4. What should be the distribution coefficient of the catalyst at the decomposition temperature? (less that 1)
- 5. How can one control the diameter of the droplet and thereby control the diameter of the nanowire? (by maintaining small equilibrium vapour pressure of the catalyst over liquid droplet).
- 6. How is the diameter of the nanowire influenced by wetting characteristic of the catalyst? (small wetting angle results in large diameter)
- 7. What is Kelvin equation?  $(\ln(p/po) = (-2\gamma\Omega)/(kTr))$
- 8. What does Kelvin equation signify? (It shows dependence of equilibrium vapour pressure or solubility on surface energy and radius of the surface)
- 9. Supersaturation of the material with catalyst should be kept low. Why? (for growth of uniform and high quality nanowire)
- 10. What will happen if supersaturation is further increased? (leads to termination of growth)
- 11. At what temperature on should carry out the synthesis of nanowires? (above the eutectic temperature of the catalyst and reactant)
- 12. Which type of materials is used as catalyst? (inert)
- 13. State the factors on which the morphology of ZnO (shown in the lecture) depends. (source, substrate temperature, air flow rate, coating thickness)
- 14. What type of morphology for ZnO was achieved in the high T zone? (nanowires)
- 15. In which temperature zone, nanorods of ZnO formed? (low T zone)

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## Solution:

- 1. liquid droplet
- 2. Interface between the substrate and liquid
- 3. 1D growth
- 4. less that 1
- 5. by maintaining small equilibrium vapour pressure of the catalyst over liquid droplet
- 6. small wetting angle results in large diameter
- 7.  $\ln(p/po) = (-2\gamma\Omega)/(kTr)$
- 8. It shows dependence of equilibrium vapour pressure or solubility on surface energy and radius of the surface
- 9. for growth of uniform and high quality nanowire
- 10. leads to termination of growth
- 11. above the eutectic temperature of the catalyst and reactant
- 12. inert
- 13. source, substrate temperature, air flow rate, coating thickness
- 14. nanowires
- 15. low T zone