

Solar Energy: The Sun to Earth Transaction

Learning objectives:

- 1) To calculate the energy received by the Earth from the Sun
- 2) To compare the energy received by Earth from the Sun, with the energy usage by humankind

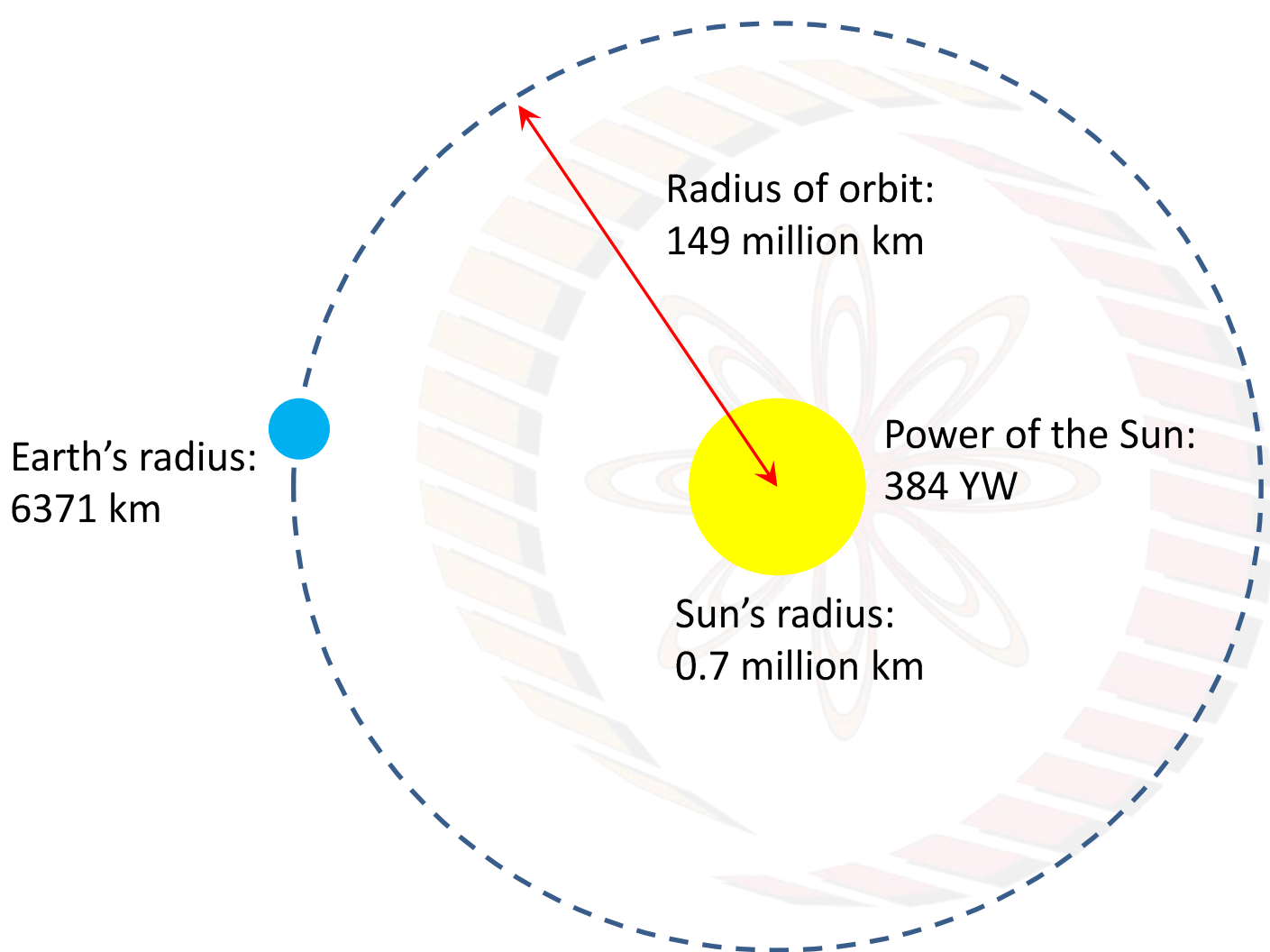
Surface of the Sun $\sim 5500\text{ }^{\circ}\text{C}$

Core of the Sun, several million $^{\circ}\text{C}$

Sun gives out **384 Yotta Watts**

$$= 384 \times 10^{24} \text{ W}$$

$$= 3.84 \times 10^{26} \text{ W}$$

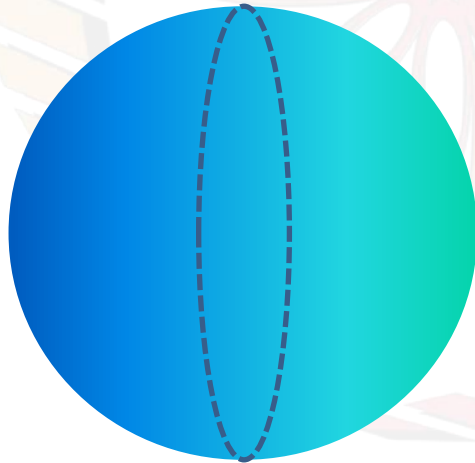


Intensity of Sun's radiation at Earth's orbit:

$$= \frac{3.84 \times 10^{26}}{4 \times 3.14 \times (1.49 \times 10^{11})^2} = 1377 \text{ W/m}^2$$

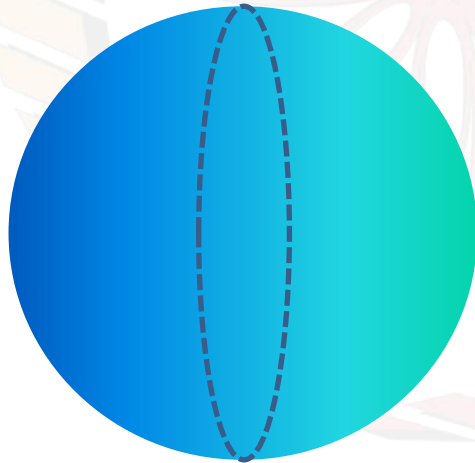
Area of Earth's disc:

$$= 3.14 \times (6.371 \times 10^6)^2 = 1.27 \times 10^{14} \text{ m}^2$$



Power received from the Sun, by Earth:

$$= 1.27 \times 10^{14} \times 1377 = 1.755 \times 10^{17} \text{ W or J/s}$$



Energy received from the Sun, by Earth each year:

$$= 1.755 \times 10^{17} \times 60 \times 60 \times 24 \times 365$$

$$= 5.5 \times 10^{24} \text{ J}$$

= 5.5 million Exa Joules per year



Humankind uses:

= 500 Exa Joules per year

Earth receives from the Sun:

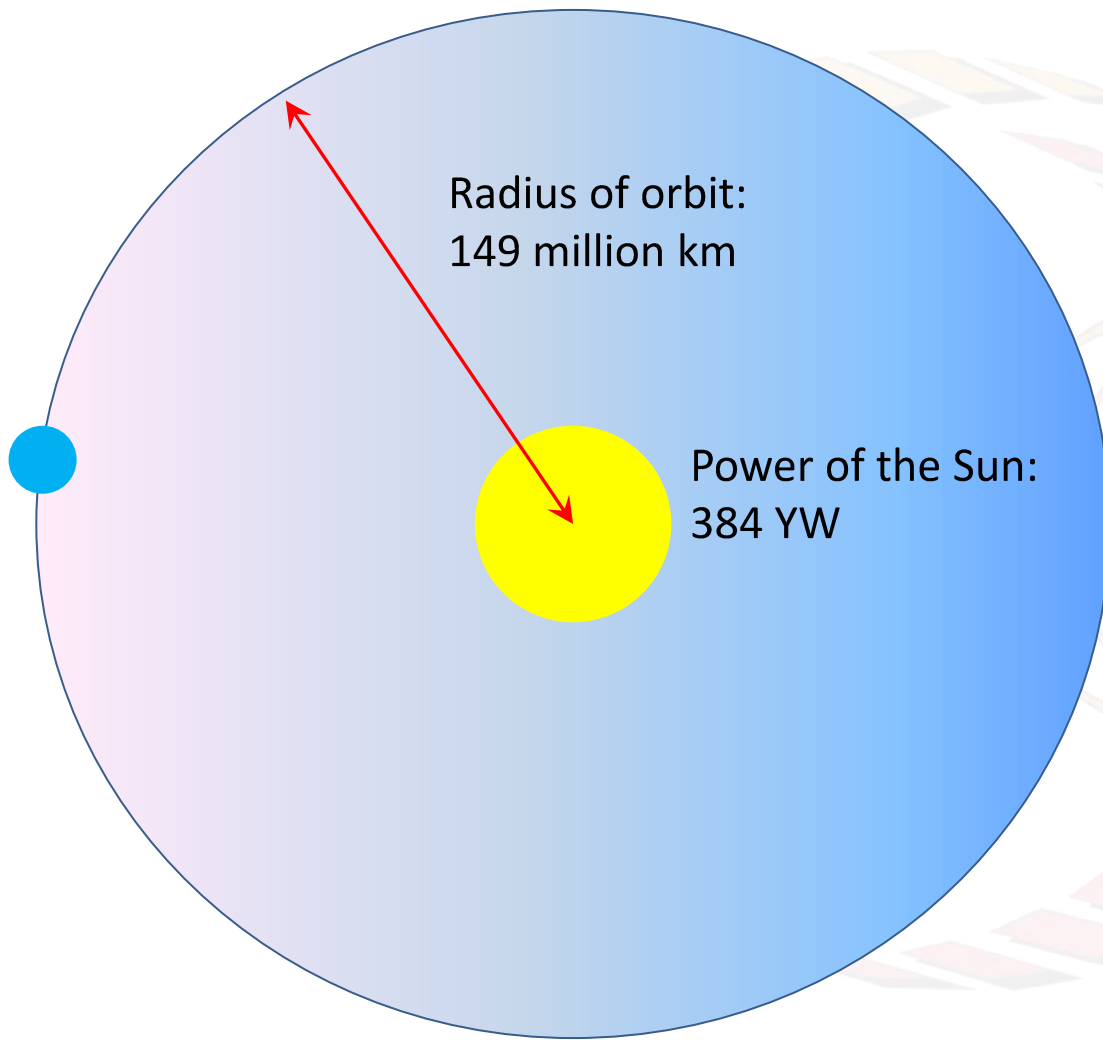
5.5 million Exa Joules per year

This is received in :

$500 / 5.5 \times 10^6 = 9 \times 10^{-5} \text{ years} = 0.033 \text{ days} = 0.79 \text{ hours}$

Since 30% of the incident energy is reflected back, on the surface of the Earth, the energy used by humankind each year is received in :

$$= 0.79/0.70 \sim \mathbf{1 \text{ hour}}$$



Dyson Sphere:

Freeman Dyson (1960)

Kardashev Scale (1964)


Type 1: All energy reaching planet

Type 2: All energy of Star

Type 3: All energy of Galaxy

Conclusions:

- 1) Earth receives nearly 5.5 million exajoules of energy from the Sun each year
- 2) The entire energy used by humankind each year, is received on the surface of the earth, from the sun, each hour!

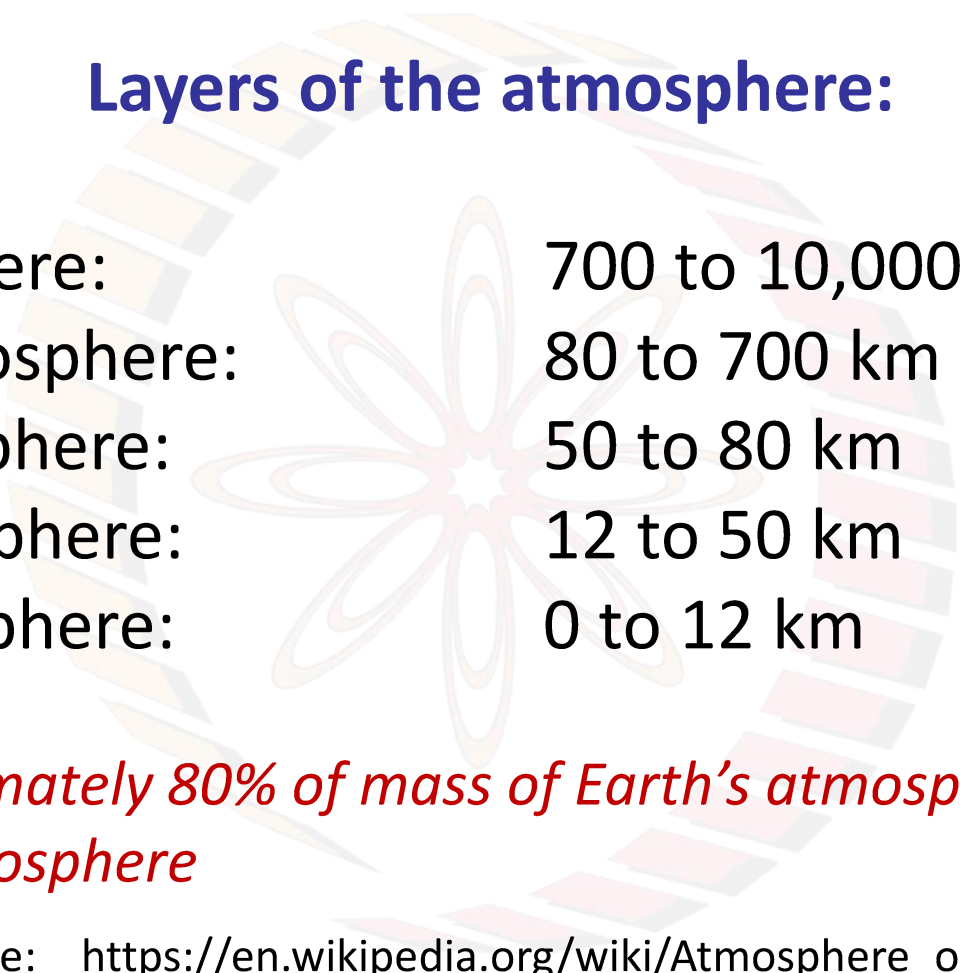


The Solar Energy Budget

Learning objectives:

- 1) To know how incoming solar energy is distributed across various phenomena on earth
- 2) To become aware of geographical and seasonal aspects associated with solar energy
- 3) To become aware of impact of time of day on ability to receive solar energy

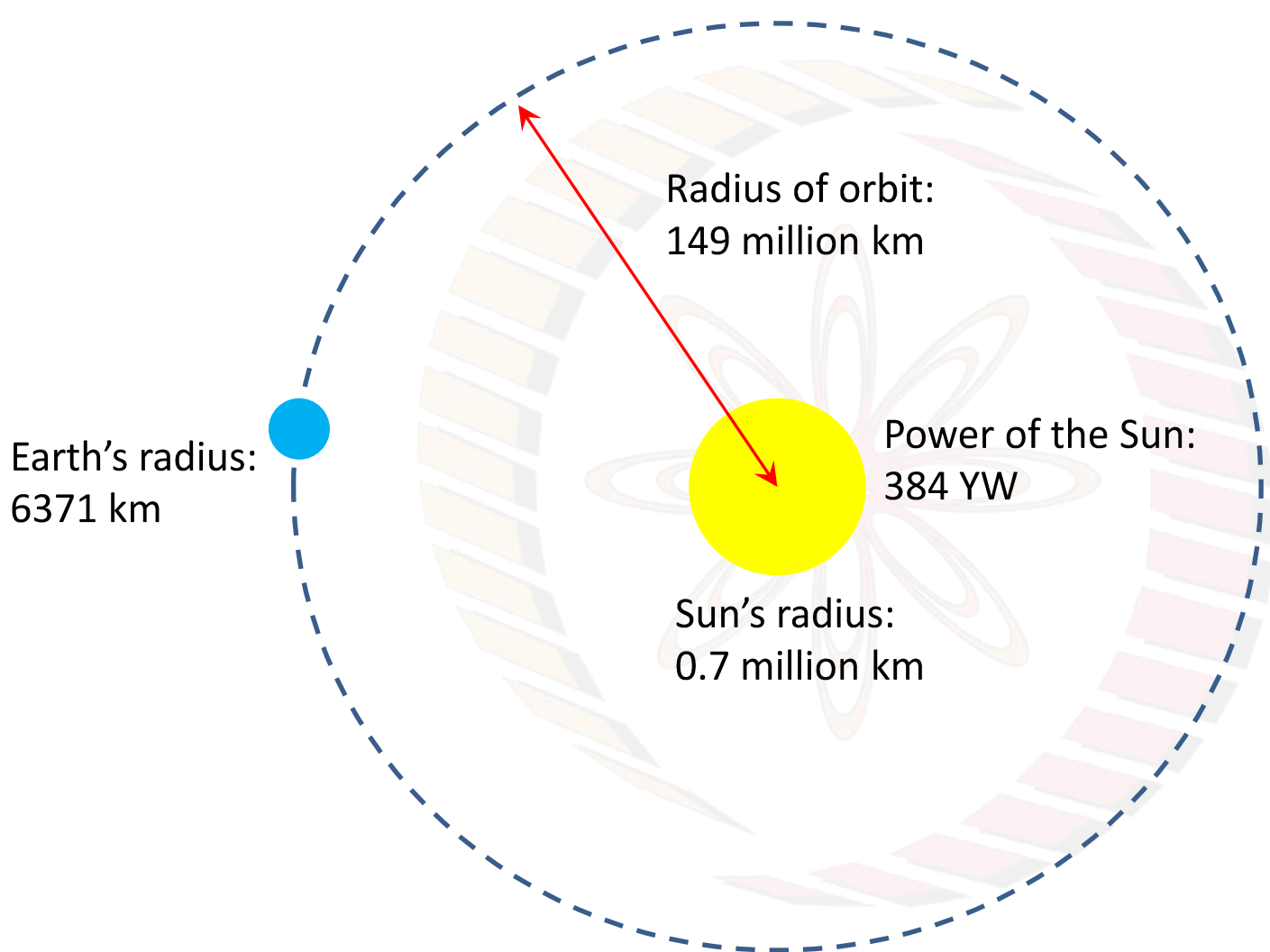
Layers of the atmosphere:



Exosphere:	700 to 10,000 km
Thermosphere:	80 to 700 km
Mesosphere:	50 to 80 km
Stratosphere:	12 to 50 km
Troposphere:	0 to 12 km

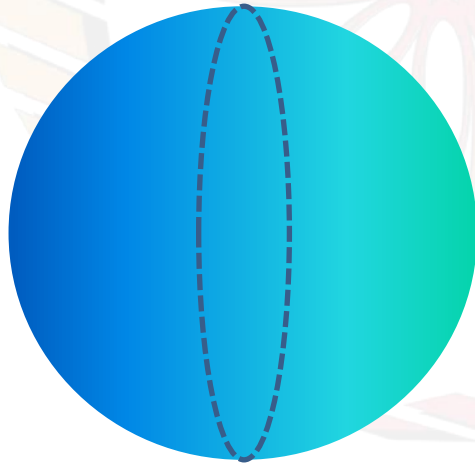
Approximately 80% of mass of Earth's atmosphere is in the Troposphere

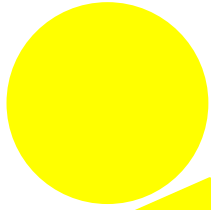
Source: https://en.wikipedia.org/wiki/Atmosphere_of_Earth



Power received from the Sun, by Earth:

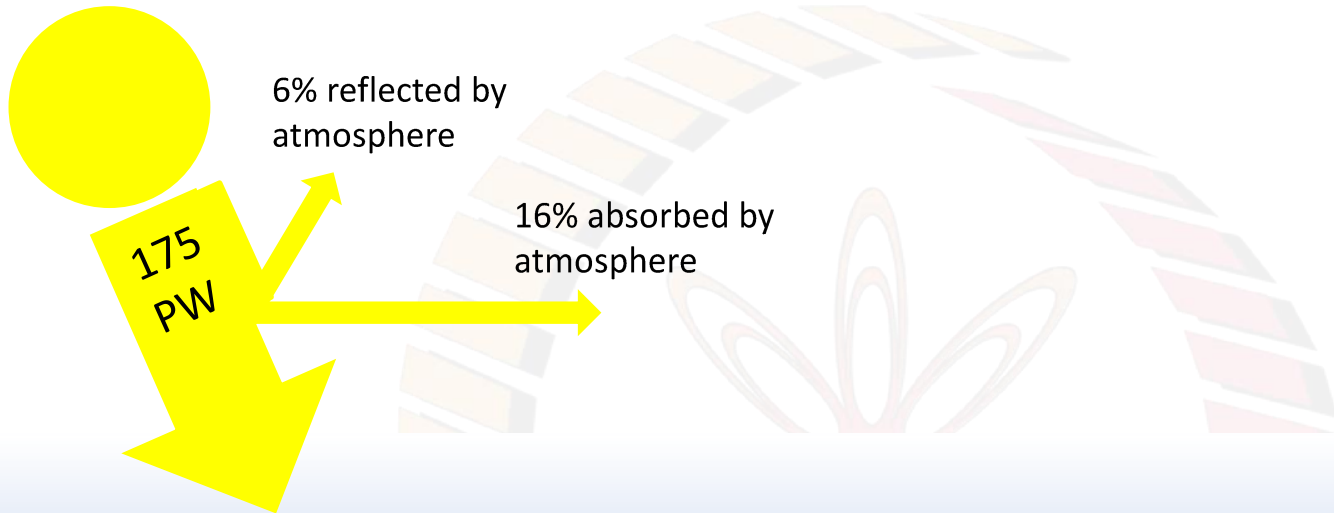
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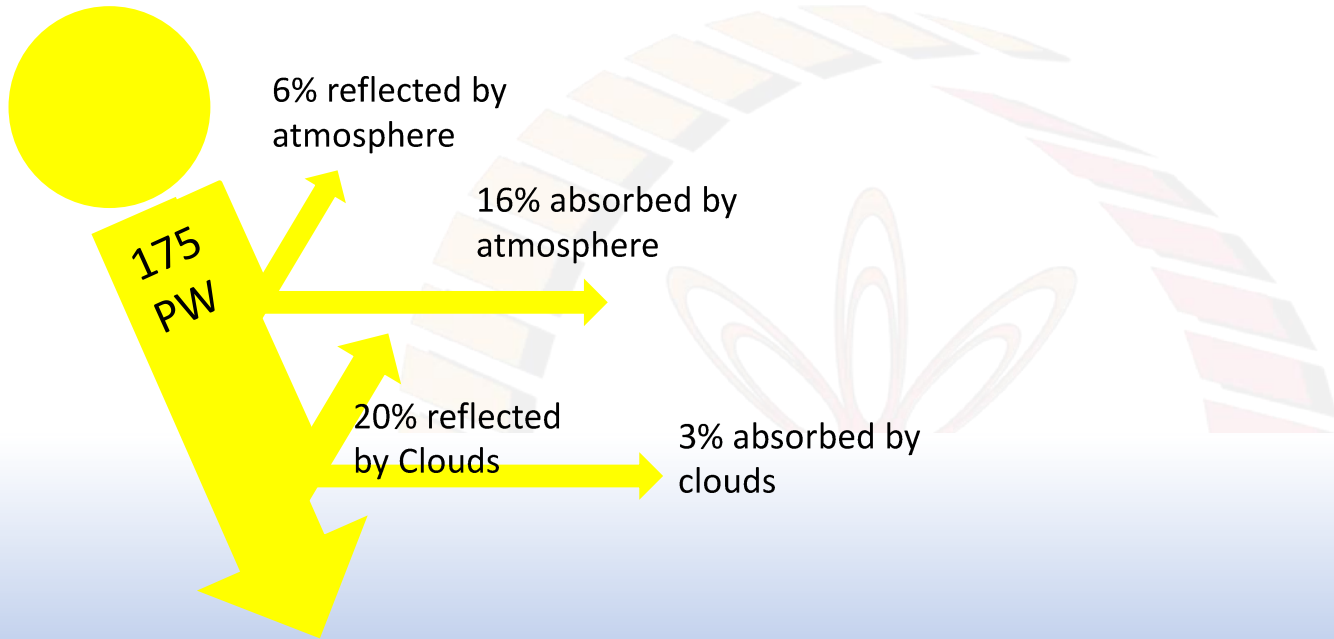


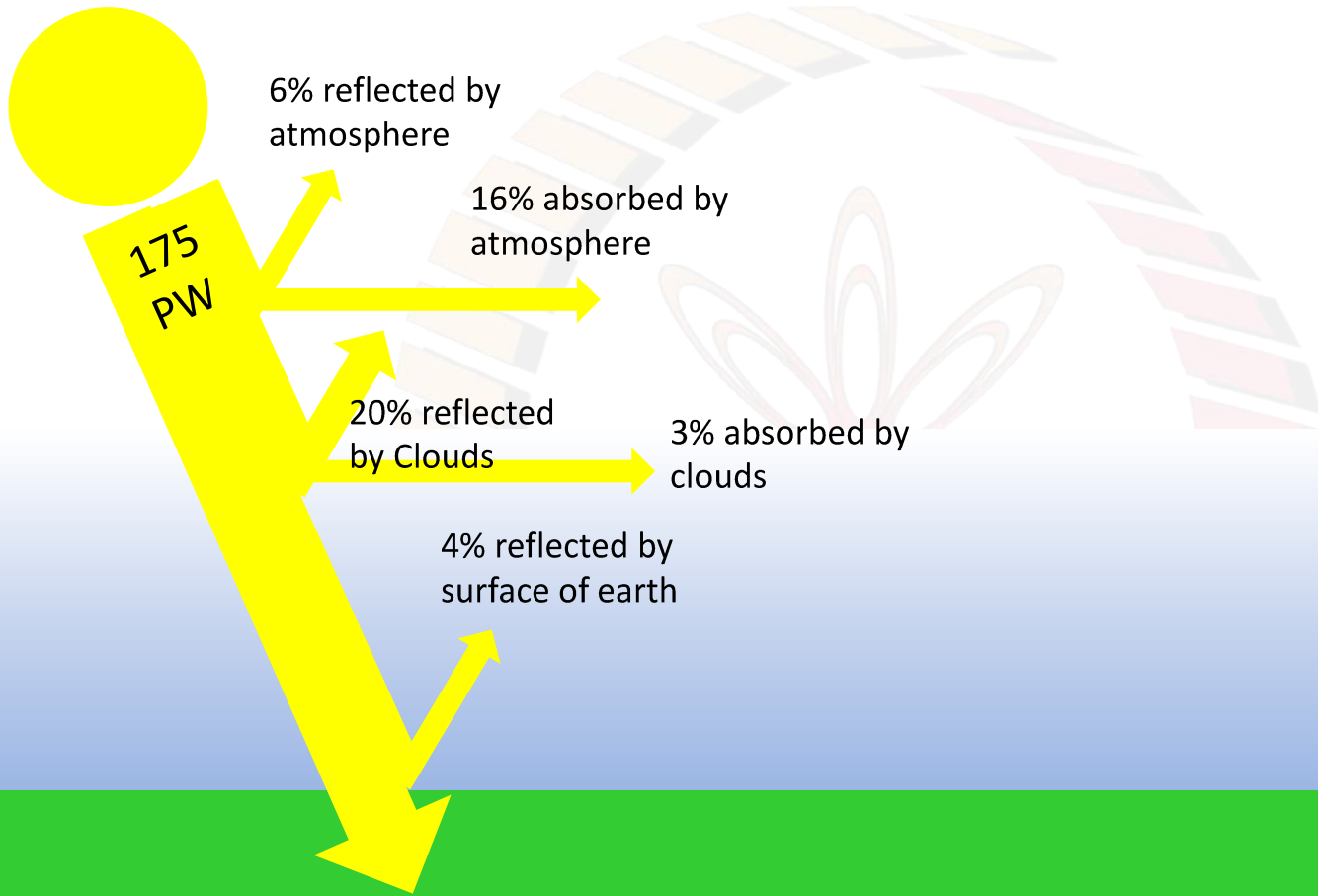


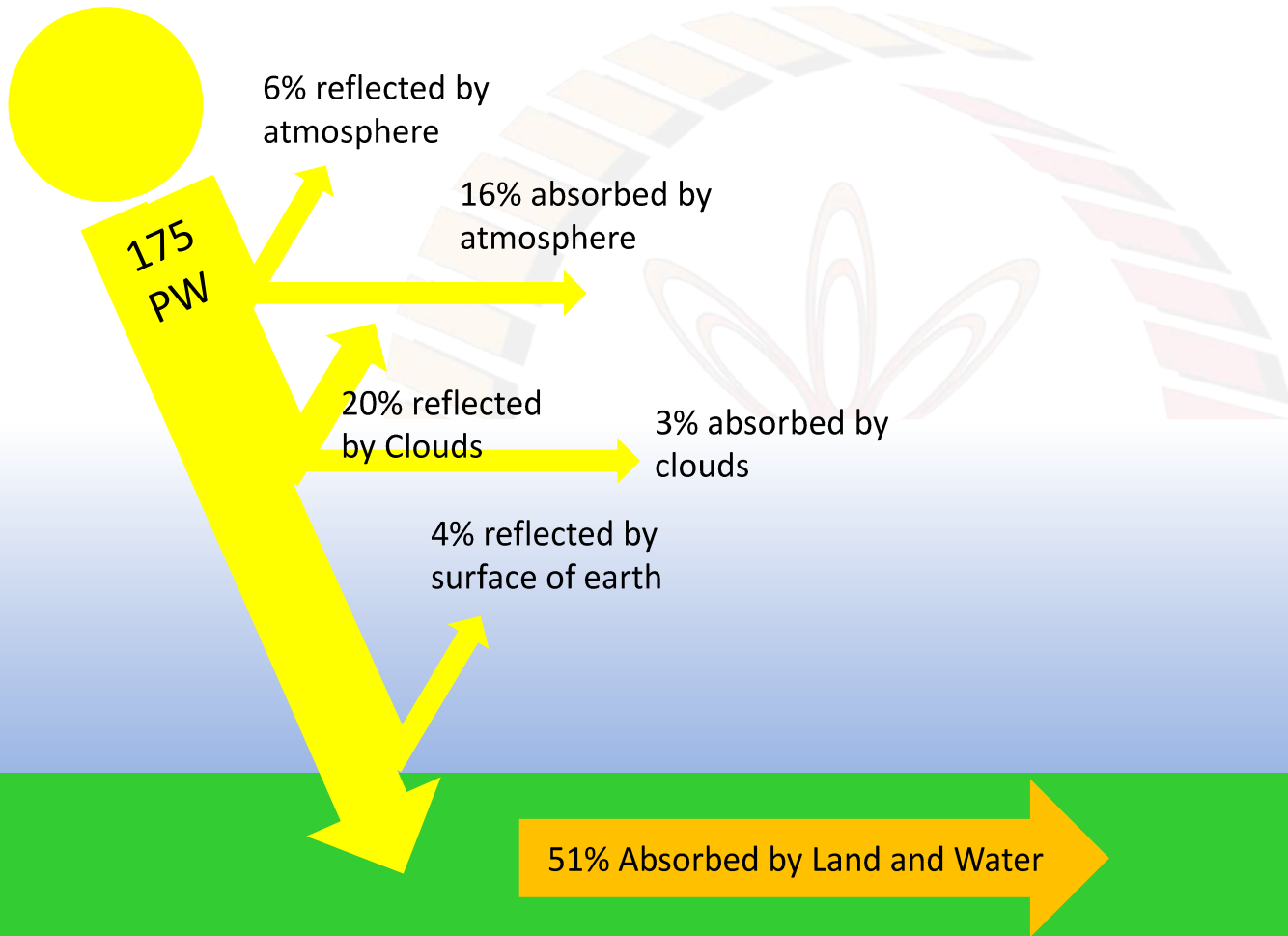
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PW

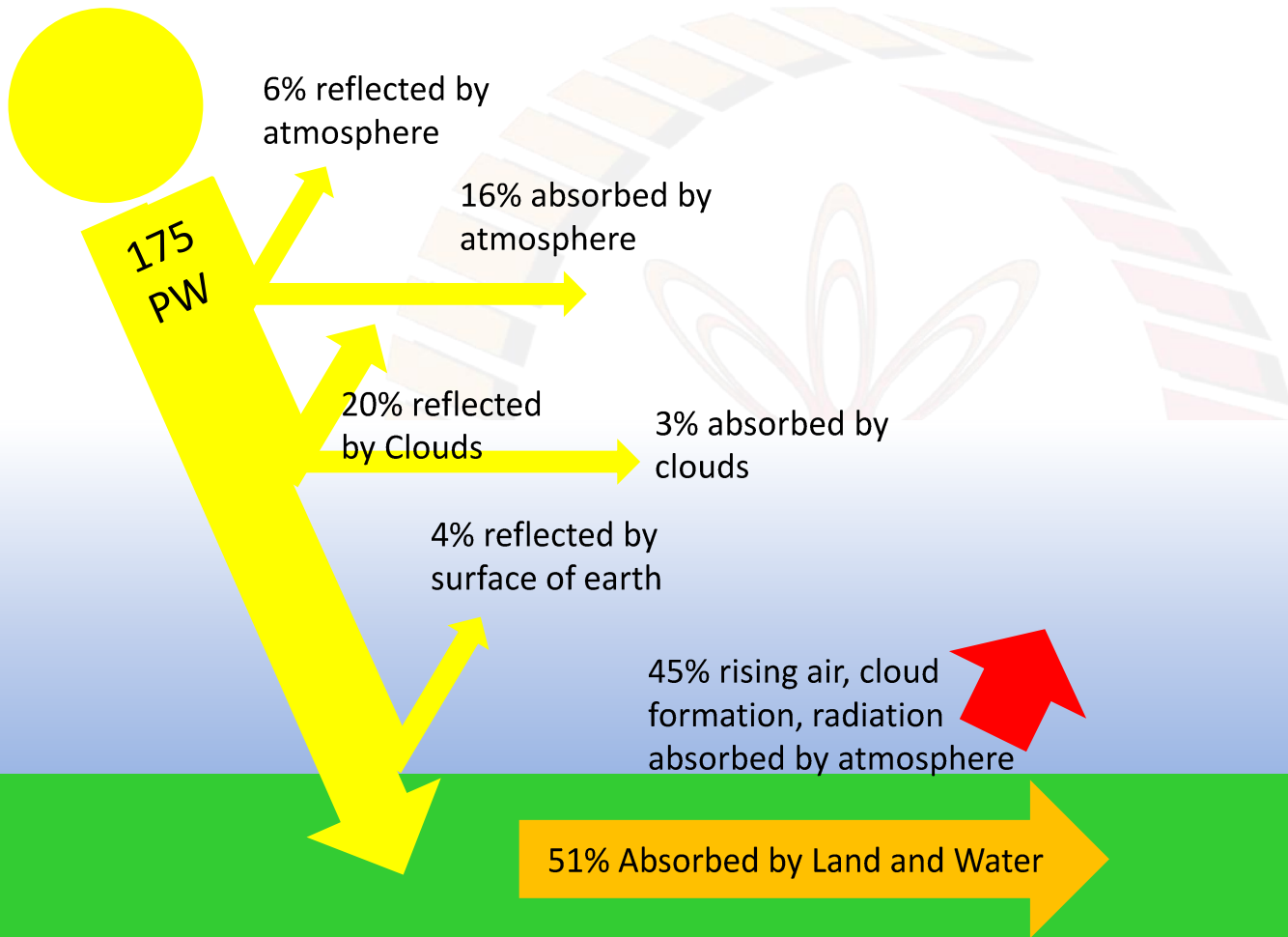


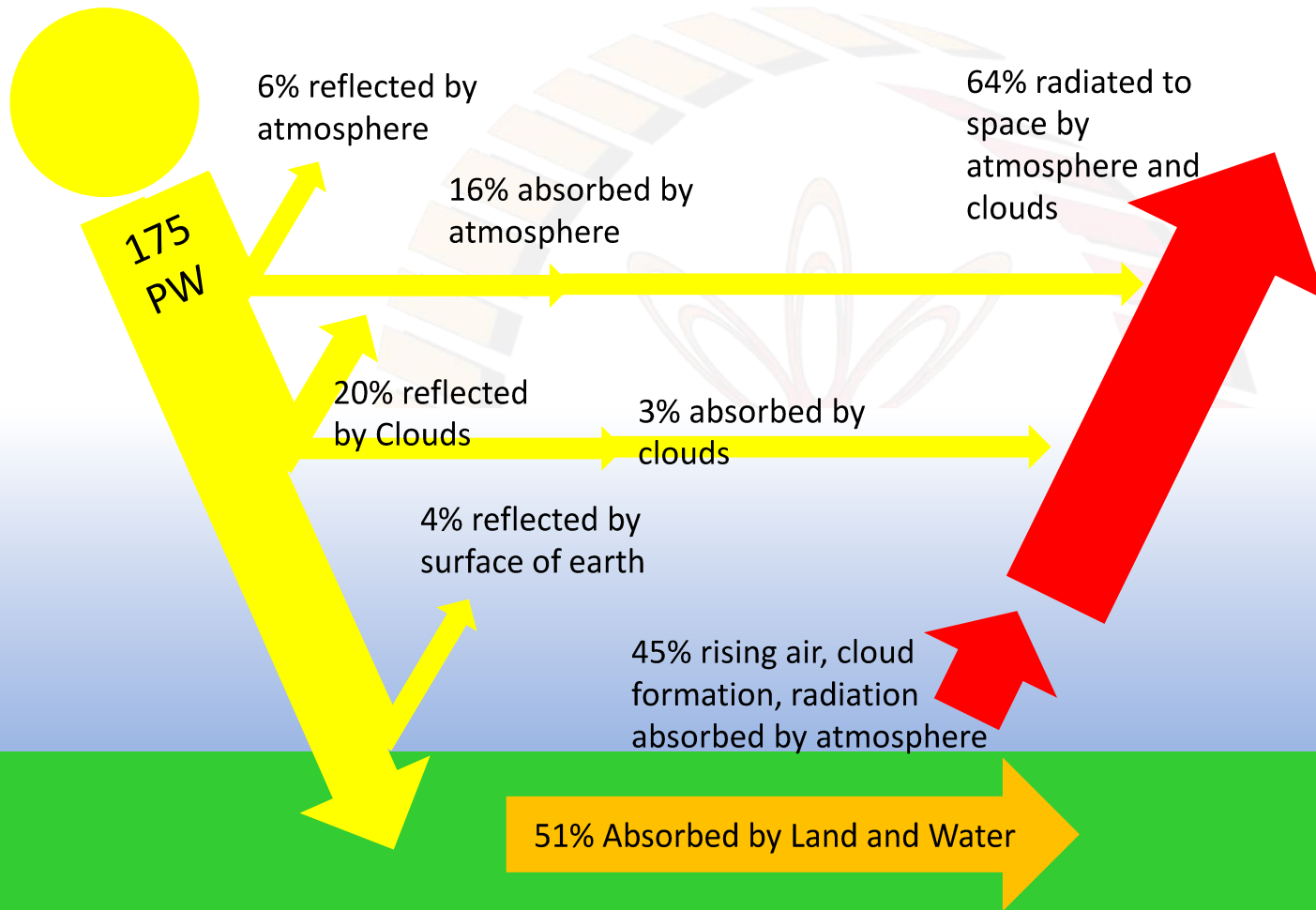


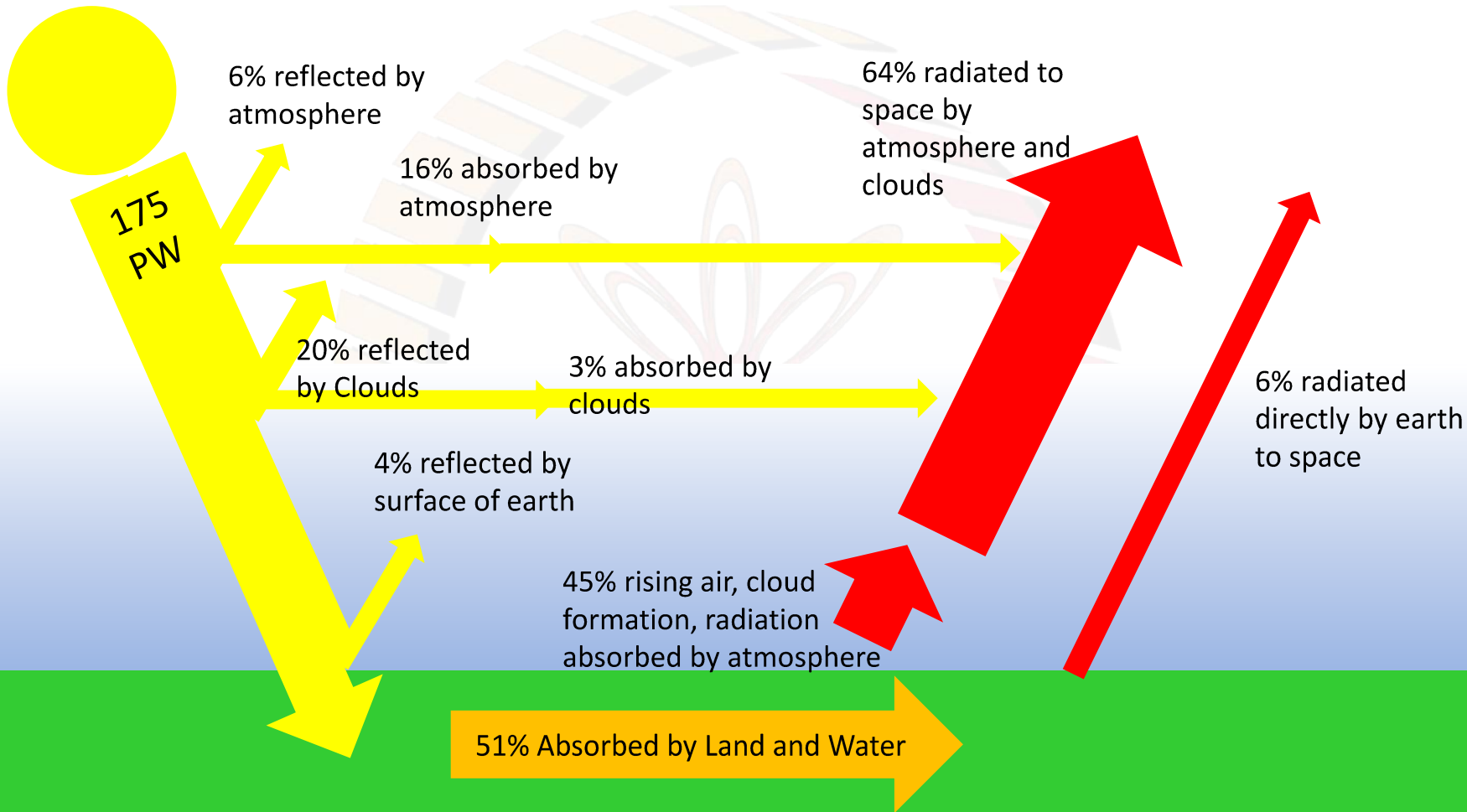


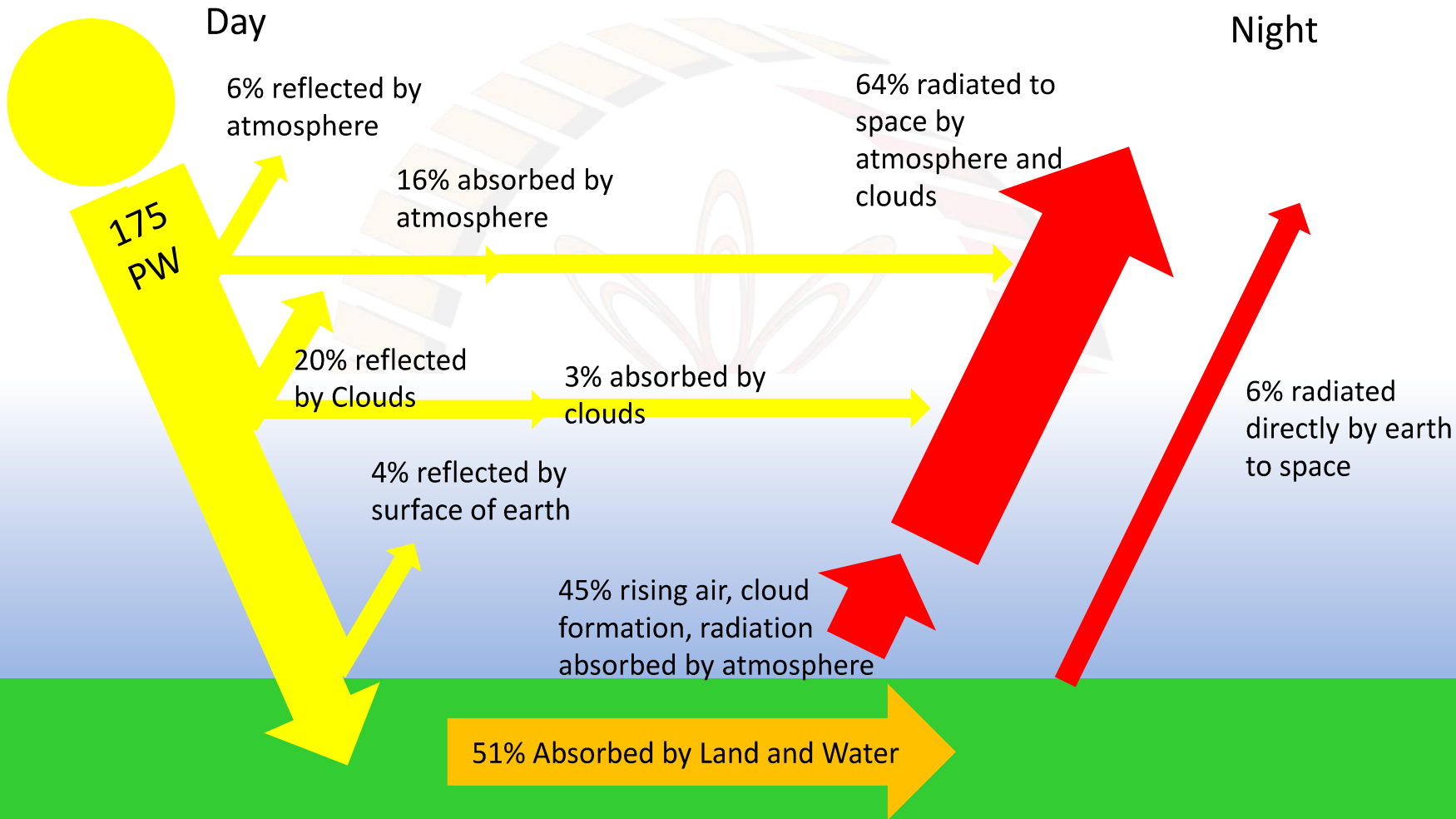


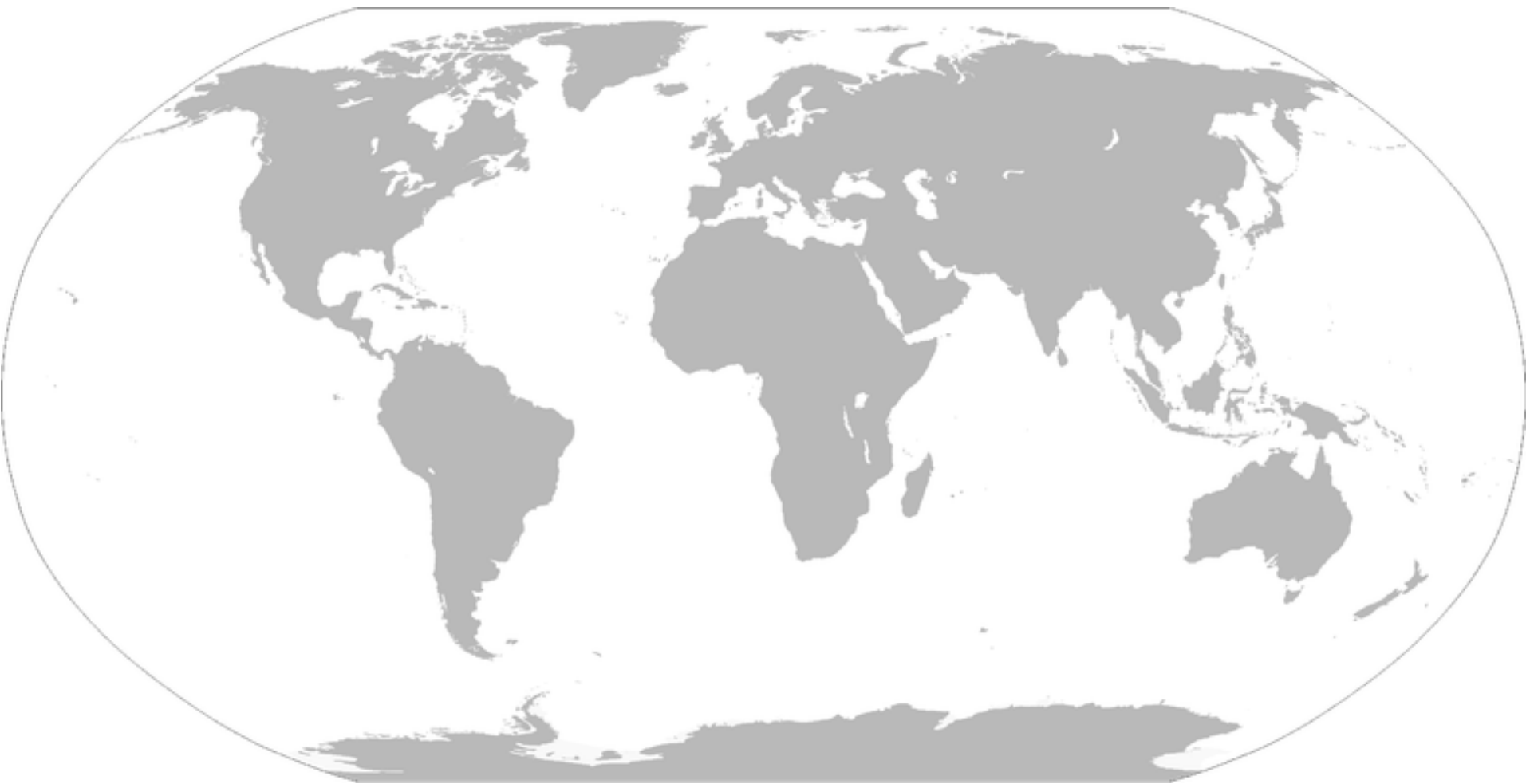






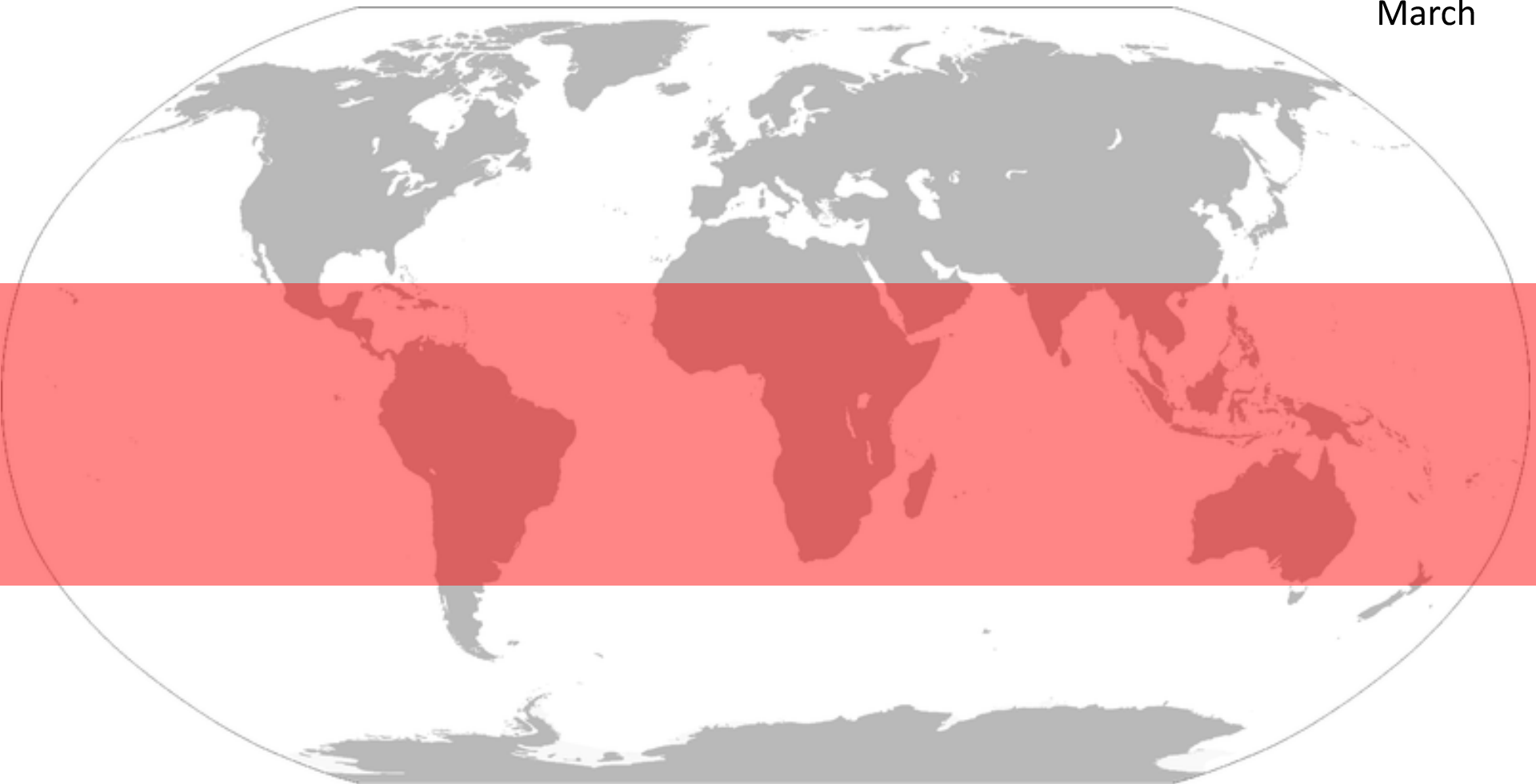






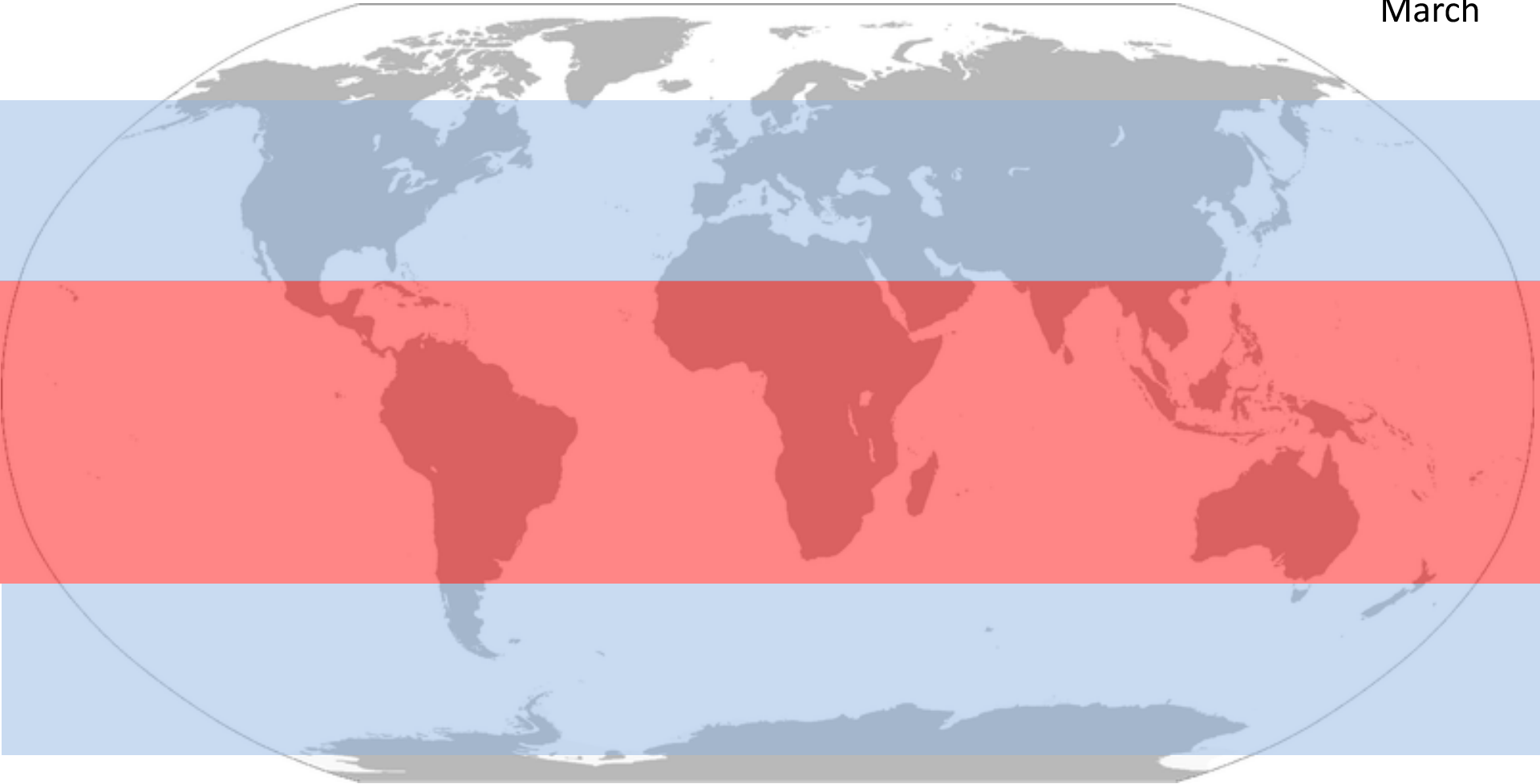
<https://en.wikipedia.org/wiki/File:BlankMap-World-large-noborders.png>

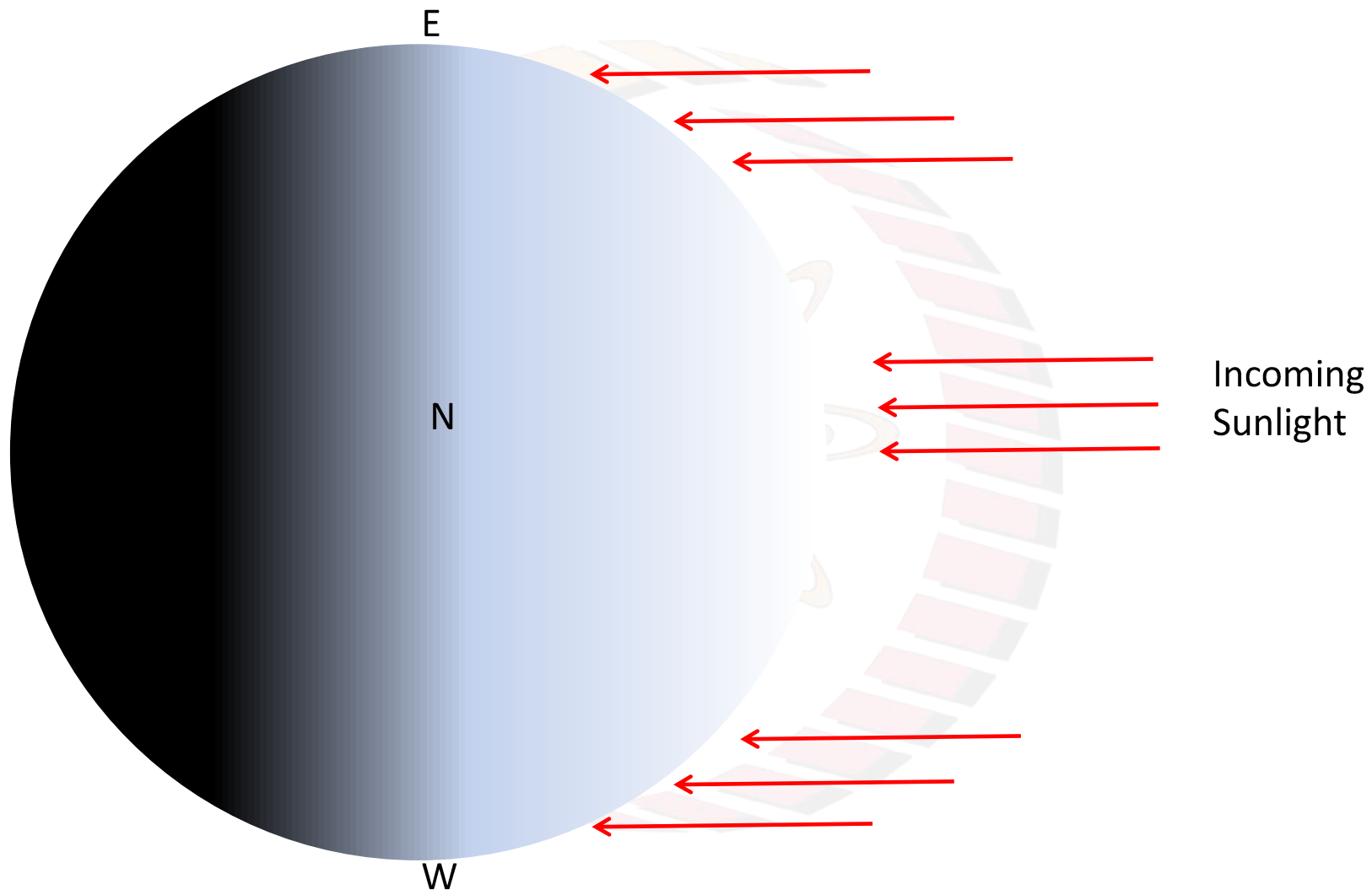
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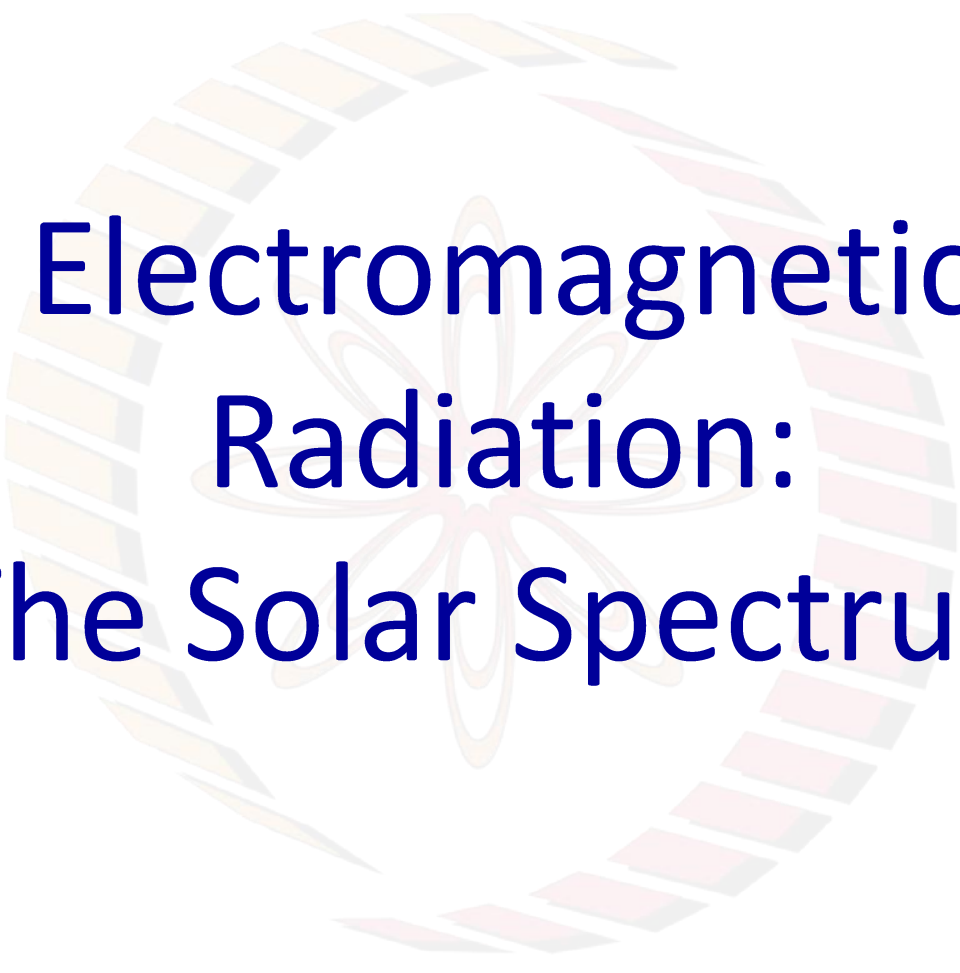
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Conclusions:

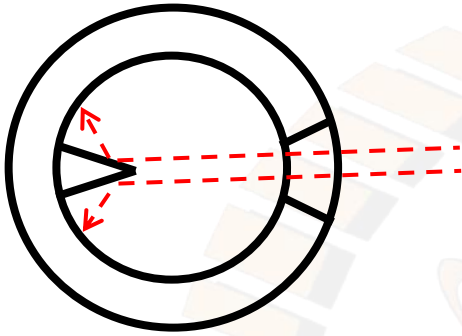
- 1) Solar energy is absorbed and released through a wide range of phenomena on earth
- 2) Geographical location and seasons are important aspects impacting solar energy received by specific locations
- 3) Time of the day is an important parameter impacting the intensity of solar energy received



Electromagnetic Radiation: The Solar Spectrum

Learning objectives:

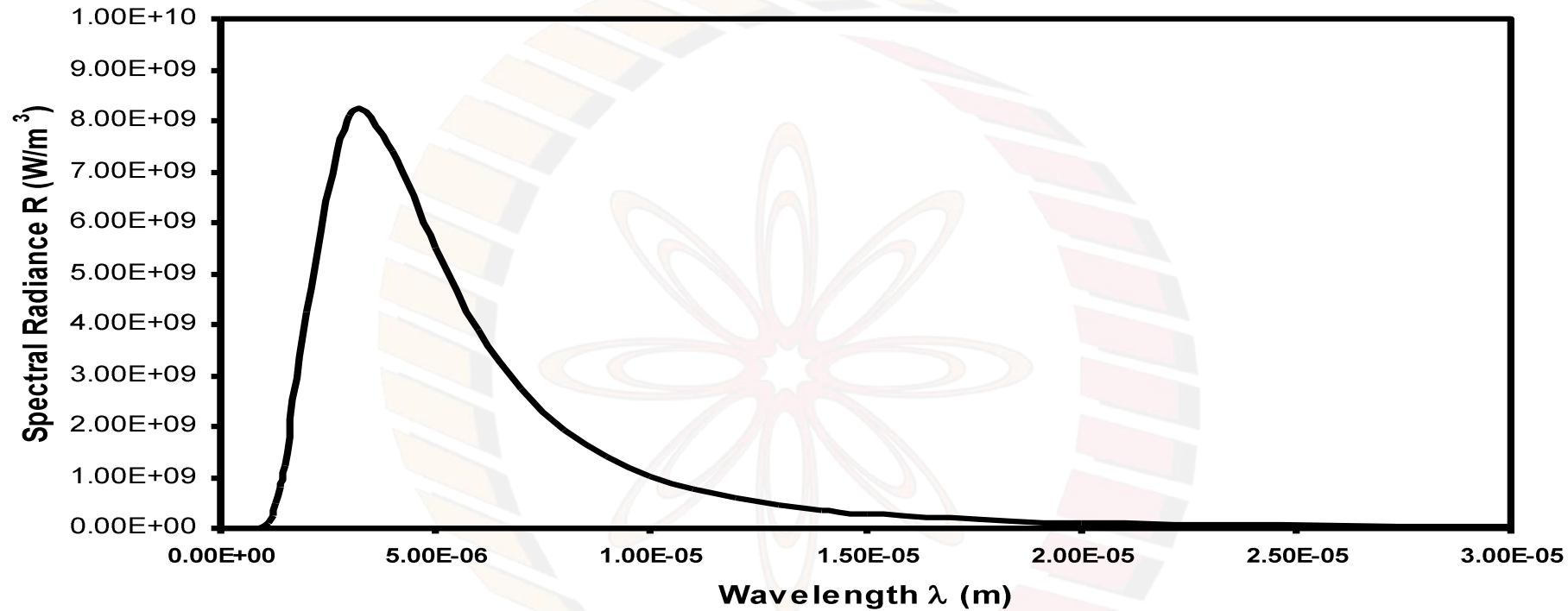
- 1) Features of electromagnetic radiation
- 2) Features of the solar energy spectrum
- 3) The ability of plants to capture visible spectrum



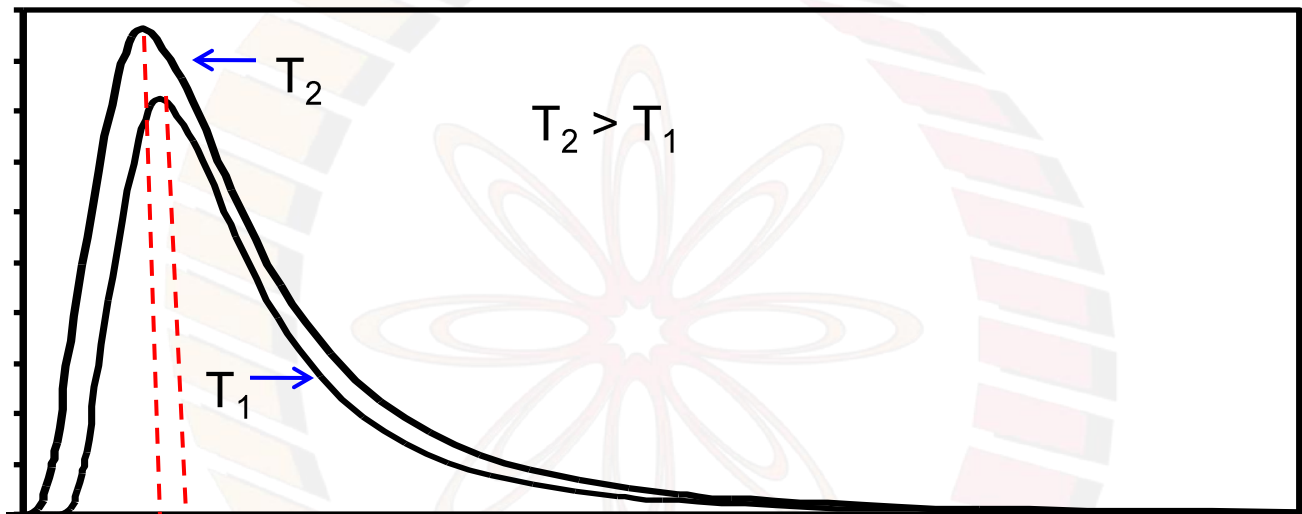
Kirchoff designed a black body in 1859

Known properties of black body radiation:

- 1) As temperature T of the body increases, intensity of the radiation from the body also increases**
- 2) Higher the temperature, lower is the wavelength of the most intense part of the spectrum.**



Black body Radiation



Wavelength (λ)

Black body Radiation

Value for 'h' obtained by Planck: 6.55×10^{-34} J.s

Current accepted value: 6.626×10^{-34} J.s

***"On the Law of Distribution of Energy in the Normal Spectrum".
Max Planck, Annalen der Physik, vol. 4, p. 553 ff (1901).***

$$\frac{1}{2} m_e v^2 = h\nu - \phi$$

Explanation of Photoelectric effect suggests:

- 1) Electromagnetic radiation of frequency ν cannot possess any arbitrary amount of energy – it can only possess energies $h\nu$, $2h\nu$, $3h\nu$, ..., $nh\nu$
- 2) Electromagnetic radiation of frequency ν behaves as though it consists of 1, 2, 3, ..., n particles each with energy $h\nu$

These 'particles' of light came to be known as 'photons'

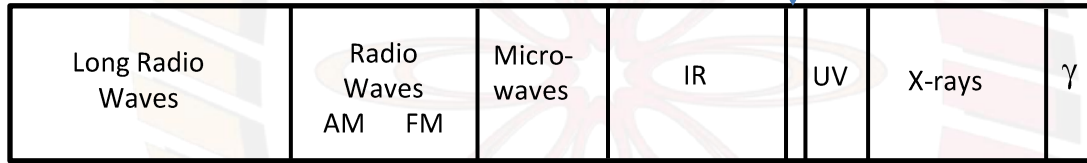
1.00E-15 1.00E-13 1.00E-11 1.00E-09 1.00E-07 1.00E-05 1.00E-03 1.00E-01 1.00E+01 1.00E+03 1.00E+05 1.00E+07

Energy (eV)

1.00E-33 1.00E-30 1.00E-27 1.00E-24 1.00E-21 1.00E-18 1.00E-15 1.00E-12

Energy J

Visible spectrum



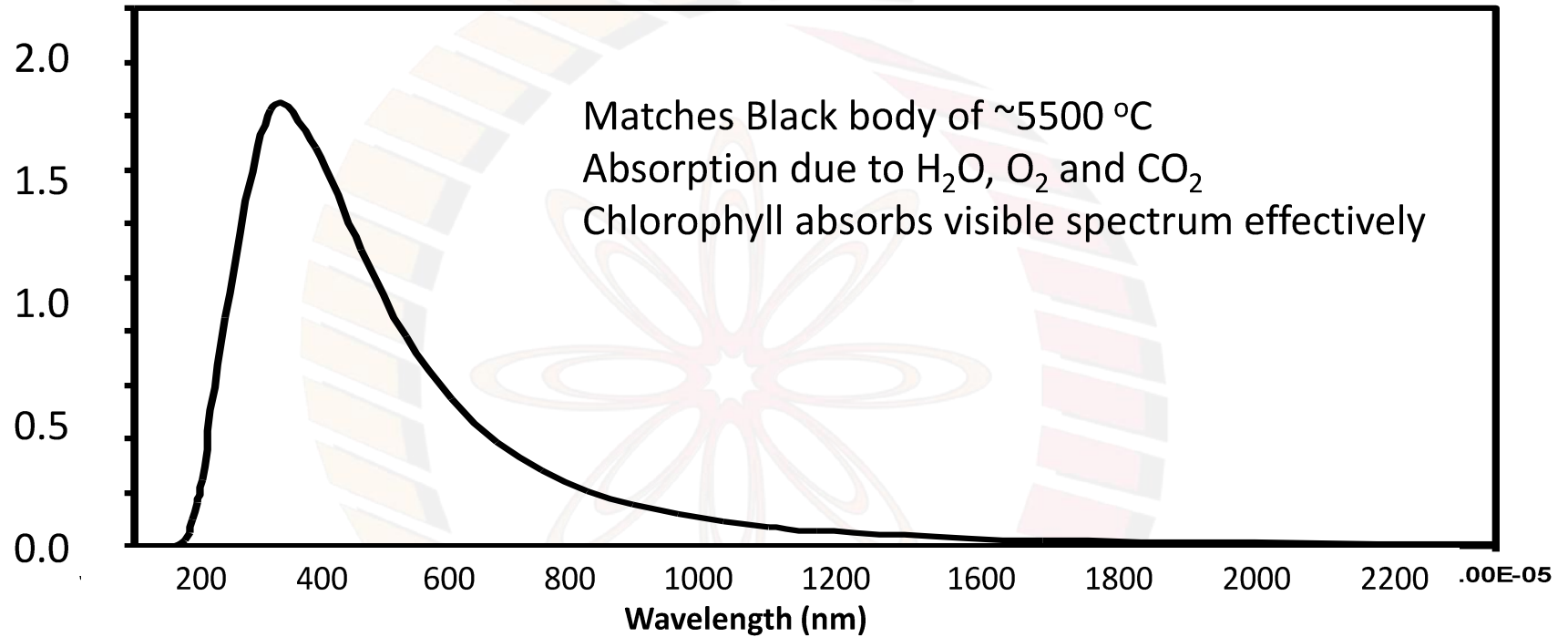
1.00E+00 1.00E+02 1.00E+04 1.00E+06 1.00E+08 1.00E+10 1.00E+12 1.00E+14 1.00E+16 1.00E+18 1.00E+20 1.00E+22

Frequency (Hz)

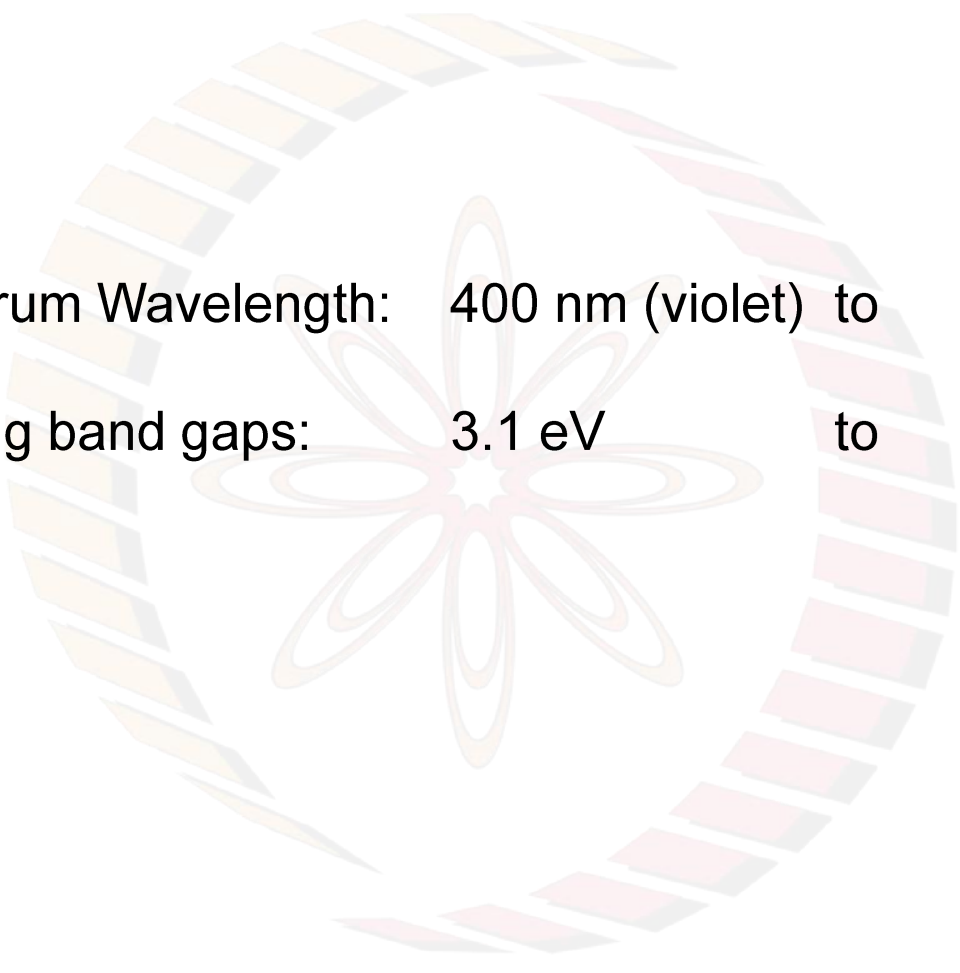
1.00E+08 1.00E+05 1.00E+02 1.00E-01 1.00E-04 1.00E-07 1.00E-10 1.00E-13

Wavelength (m)

Adapted from: https://en.wikipedia.org/wiki/Electromagnetic_radiation



Solar Radiation



Visible Spectrum Wavelength: 400 nm (violet) to 700 nm (red)

Corresponding band gaps: 3.1 eV to 1.8 eV

Conclusions:

- 1) Solar energy spectrum is consistent with black body radiation
- 2) Visible spectrum a very small fraction of electromagnetic radiation
- 3) Chlorophyll ideally suited for absorbing visible spectrum